```
=> d que
               STR
```

VAR G1=AK/CB NODE ATTRIBUTES: NSPEC IS R AT 1 NSPEC IS R AT 2 NSPEC IS R AT 3 DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 4

STEREO ATTRIBUTES: NONE

L5 33032 SEA FILE=REGISTRY SSS FUL L3 L10 STR



VAR G1=AK/CB REP G2=(1-20) 9-6 11-8 NODE ATTRIBUTES: NSPEC IS R AT 1 NSPEC IS R AT 2 NSPEC IS R AT 3 DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 11

STEREO ATTRIBUTES: NONE

167 SEA FILE=REGISTRY SUB=L5 SSS FUL L10 L13 96 SEA FILE=REGISTRY ABB=ON PLU=ON L12 NOT 1-100/X 51 SEA FILE-HCAPLUS ABB-ON PLU-ON L13 L14

L15 41 SEA FILE=HCAPLUS ABB=ON PLU=ON L14 AND (1840-2004)/PRY, AY

, PY

L16	1	SEA	FILE=HCAPLUS	ABB=ON	PLU=ON	L15 AND	ELECTROCHEM?
L17	7	SEA	FILE=HCAPLUS	ABB=ON	PLU=ON	L15 AND	ELECTRO?
L18	41	SEA	FILE=HCAPLUS	ABB=ON	PLU=ON	(L15 OR	L16 OR L17)
L21	3	SEA	FILE=HCAPLUS	ABB=ON	PLU=ON	L15 AND	ELECTRO?/SC,SX
1.22	41	SEA	FILE-HCAPLUS	ARR=ON	PLH=ON	1.18 OR	1.21

=> d 122 1-41 ibib ed abs hitstr hitind

L22 ANSWER 1 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:380773 HCAPLUS Full-text

DOCUMENT NUMBER: 144:413589

TITLE: Composition containing unsaturated silicone

compounds, and uses as dental materials and others

INVENTOR(S): Zech, Joachim; Bissinger, Peter

PATENT ASSIGNEE(S): 3M Espe A.-G., Germany SOURCE: Eur. Pat. Appl., 26 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent

LANGUAGE: English FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PA	PATENT NO.				KIND DATE			APPLICATION NO.				DATE				
EP	P 1650263				A1 20060426				EP 2		1606	5	20050725			
	R:	PT,	IE,	SI,		LV,	ES, FI, YU				IT,	LI,				
EP	1652						2006	0503		EP 2			3		2	0041012
	R:	PT,		SI,			ES, FI,				IT,					
US	2006				A1		2006	0525		US 2	005-		40		2	0051011
JP	2006	1176	55		A		2006	0511		JP 2	005-	 2980: 	36		2	0051012
PRIORIT	Y APP	LN.	INFO	. :						EP 2	004-		3	i	A 2	0041012

ED Entered STN: 27 Apr 2006

AB A curable composition comprises addition curable silicone component (A) having a linear siloxane backbone at least one unsatd., non-activated side group pendant from the Si atom or atoms in the backbone, component (A) having an equivalent weight in the range of about 68 to about 1000, crosslinker component (B) comprising at least 2 Si-H groups, catalyst component (C) being able to catalyze the reaction between component (A) and component (B), filler component (D), wherein the equivalent weight is defined as (mol. weight of the mol.) / (number of unsaturations in the mol.), wherein the total number of carbon-carbon double bonds in component (A) is at least 3, and wherein the unsatd., non-activated side group does not comprise a -O-Si-CH=CH2 moiety.

(composition containing unsatd. silicone compds., and uses as dental materials and others)

RN 884488-01-1 HCAPLUS

CN Cyclotetrasiloxane, 2,2'-[(1-methylethylidene)bis(4,1-phenyleneoxy-2,1-ethanediyloxy-3,1-propanediyl)]bis[2,4,6,8-tetramethyl- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

CC 37-3 (Plastics Manufacture and Processing) Section cross-reference(s): 63

IT 100-42-5DP, Styrene, reaction products with polymethylsiloxane 592-42-7DP, 1,5-Hexadiene, reaction products with polymethylsiloxane 884488-00-DP 994488-01-1P

(composition containing unsatd. silicone compds., and uses as dental materials and others)

REFERENCE COUNT: 7

THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L22 ANSWER 2 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:216257 HCAPLUS Full-text DOCUMENT NUMBER: 144:257284

DOCUMENT NUMBER: 144:25728
TITLE: Nonagueou

IIILE: Nonaqueous electrolyte solution

containing polysiloxanes and its use in battery INVENTOR(S): Kashida, Osamu; Miyawaki, Satoru; Ichinohe, Shoji;

Aramata, Mikio

PATENT ASSIGNEE(S): Shin-Etsu Chemical Industry Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 9 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent
LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
JP 2006066095	A	20060309	JP 2004-243982	20040824		
			<			
PRIORITY APPLN. INFO.:			JP 2004-243982	20040824		

OTHER SOURCE(S): MARPAT 144:257284

ED Entered STN: 10 Mar 2006

- AB The solution contains nonaq. solvents, electrolyte salts (e.g., Li salt), and cyclic siloxanes I (RI = H, Me; R2 = Me, Et, Pr, acetyl; m = 3-6; n = 1-6). The battery using the solution shows good temperature performance and highoutput performance.
- IT 131718-86-0 877420-97-8 677421-00-6
 (nonag. electrolyte solution containing polyoxyethylene-containing
- cyclic siloxanes for battery) RN 131718-86-0 HCAPLUS
- CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-tetrakis(4,7,10,13-tetraoxatetradec-1-yl)- (9CI) (CA INDEX NAME)

- (CH2) 3-0-CH2-CH2-O-CH2-CH2-O-CH2-OH2-OH2-OMe
- RN 877420-97-8 HCAPLUS
- CN Cyclotetrasiloxane, 2,4,6,8-tetrakis[3-[2-(2-methoxyethoxy)ethoxy]-2-methylpropyl]-2,4,6,8-tetramethyl- (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

- RN 877421-00-6 HCAPLUS
- CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-tetrakis(2-methyl-4,7,10,13-tetraoxatetradec-1-yl)- (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

_____Me

IT 17232-95-0P

(nonaq. electrolyte solution containing polyoxyethylene-containing cvclic siloxanes for battery)

RN 17232-95-0 HCAPLUS

CN Cyclotetrasiloxane, 2,4,6,8-tetrakis[3-[2-(2-

methoxyethoxy]propy1]-2,4,6,8-tetramethy1- (CA INDEX NAME)

PAGE 1-B

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

Section cross-reference(s): 37

- ST nonaq electrolyte soln polyoxyethylene cyclosiloxane battery
- IT Battery electrolytes

(nonaq. electrolyte solution containing polyoxyethylene-containing cyclic siloxanes for battery)

Cyclosiloxanes

(nonaq. electrolyte solution containing polyoxyethylene-containing cyclic siloxanes for battery)

IT 21324-40-3, Lithium hexafluorophosphate

(electrolyte; nonaq. electrolyte solution containing polyoxyethylene-containing cyclic siloxanes for battery)

II 131718-86-0 877420-97-8 877421-00-6

(nonaq. electrolyte solution containing polyoxyethylene-containing cyclic siloxanes for battery)

IT 17232-95-0P

(nonaq. electrolyte solution containing polyoxyethylene-containing cyclic siloxanes for battery)

IT 2370-88-9, 1,3,5,7-Tetramethylcyclotetrasiloxane 13752-97-1 (nonaq. electrolyte solution containing polyoxyethylene-containing cyclic siloxanes for battery)

L22 ANSWER 3 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:54848 HCAPLUS Full-text DOCUMENT NUMBER: 144:155843

TITLE:

INVENTOR(S):

Dental composition containing unsaturated

carbosilane-containing components Bissinger, Peter; Eckert, Adrian

PATENT ASSIGNEE(S): 3M Espe AG, Germany
SOURCE: PCT Int. Appl., 61 pp.
CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.					KIND DATE												
											2004-1				20040714		
	W:	CH, GB, KR, MX,	CN, GD, KZ, MZ,	CO, GE, LC, NA,	CR, GH, LK, NI,	CU, GM, LR, NO,	CZ, HR, LS, NZ,	DE, HU, LT, OM,	DK, ID, LU, PG,	DM, IL, LV, PH,	BG, DZ, IN, MA, PL,	EC, IS, MD, PT,	EE, JP, MG, RO,	EG, KE, MK, RU,	ES, KG, MN, SC,	FI, KP, MW, SD,	
	RW:	VC, AT, IE, CG, GM,	VN, BE, IT, CI, KE,	YU, BG, LU, CM, LS,	ZA, CH, MC, GA, MW,	ZM, CY, NL, GN, MZ,	ZW CZ, PL, GQ,	DE, PT, GW, SD,	DK, RO, ML,	EE, SE, MR,	ES, SI, NE, TZ,	FI, SK, SN,	FR, TR, TD,	GB, BF, TG,	GR, BJ, BW,	HU, CF, GH,	
AU	2004									AU 2		3214	29		2	0040714	
CA	2573	496			A1		2006	0119		CA 2	2004-	2573 	496		2	0040714	
EP	1765	261			A1		2007	0328	1	EP 2		7409 	71		2	0040714	
	R:										ES,				GR,	HU,	
CN	1988	875			A		2007	0627	•	CN 2		8004	3602		2	0040714	
JP	2008	5059	39		Т		2008	0228		JP 2	2007-		68		2	0040714	
US	2007	2074	43		A1		2007	0906	1	US 2	2007-	5720 	69		2	0070112	
KR	2007	0323	77		A		2007	0321	1	KR 2		7034	34		2	0070213	
RIT	Y APP	LN.	INFO	. :					1	WO 2		EP77	46	i	A 2	0040714	

ED Entered STN: 20 Jan 2006

AB The invention relates to a curable dental composition comprising (a) at least 1% by weight of a carbosilane-containing component comprising at least one Silaryl bond, at least one silicon atom, at least one unsatd. moiety, and no Si-O bond, (b) an initiator, (c) optionally at least 3% by weight of a filler, and (d) optionally a component selected from modifiers, dyes, pigments, thixotropic agents, flow improvers, polymeric thickeners, surfactants, odorous substances, diluting agent(s) and flavoring in an amount of less than 25% by weight. For example, a dental composition having opacity of 87.9% contained 1-[2, 4,6- tris[dimethyl=2-(5/6-methacroyloxybicyclo]2,2,1]hept-2-yl)ethyl]silyl]phenoxy]-3-[dimethyl[2-(5/6-methacroyloxybicyclo]2,2.1]hept-2-yl)ethyl]silyl]propane 29.%, bis[2,6-dichlorobenzoyl] (4-butylphenyl)-

phosphane oxide 0.1%, and silaned quartz (mean particle size <2 μ m) 70.0%. T 873794-35-5

11 010101 00 0

(polymerizable dental compns. containing unsatd. carbosilane

components)

RN 873794-35-5 HCAPLUS

2-Propenoic acid, 2-methyl-, (2,4,6,8-tetramethylcyclotetrasiloxane-2,4,6,8-tetrayl)tetrakis(3,1-propanediyloxy-2,1-ethanediyl) ester (9CI) (CA INDEX NAME)

PAGE 1-A

$$\begin{array}{c} \text{H2C} & \text{O} \\ \text{Me} & \text{CL} & \text{O} & \text{CH2-CH2-O-} & \text{(CH2)} & \text{3} \\ \text{Me} & \text{S1} & \text{Me} & \text{S1} \\ \text{Me} & \text{CL} & \text{O} & \text{CH2-CH2-O-} & \text{(CH2)} & \text{S1} \\ \text{Me} & \text{CL} & \text{O} & \text{CH2-CH2-O-} & \text{(CH2)} & \text{S1} \\ \text{Me} & \text{CL} & \text{O} & \text{CH2-CH2-O-} & \text{(CH2)} & \text{S1} \\ \text{Me} & \text{CL} & \text{O} & \text{CH2-CH2-O-} & \text{CH2-CH2-O-} \\ \end{array}$$

PAGE 1-B

ICM A61K006-093

63-7 (Pharmaceuticals)

Section cross-reference(s): 37

281-50-5D, Hydrogenoctasilsesquioxane, methacroyloxy derivs. 873794-35-5 873794-36-6

(polymerizable dental compns. containing unsatd. carbosilane components)

REFERENCE COUNT: THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L22 ANSWER 4 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2005:1307898 HCAPLUS Full-text DOCUMENT NUMBER: 144:43213

TITLE: Ion-conductive rubber rolls showing stable resistivity against voltage change and their

compositions

INVENTOR(S): Ikeno, Masayuki; Sudo, Shigeki

PATENT ASSIGNEE(S): Shin-Etsu Chemical Industry Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 18 pp.

CODEN: JKXXAF DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2005344102	A	20051215	JP 2005-125207	20050422
			<	
CN 1693370	A	20051109	CN 2005-10076270	20050430
			<	
PRIORITY APPLN. INFO.:			JP 2004-139167 A	20040507
			/	

ED Entered STN: 15 Dec 2005

Compns. of (A) ≥ 2 (/mol.) Si-bonded alkenyl-containing organopolysiloxanes, (B) organohydrogen polysiloxanes having ≥ 2 SiH groups and polyethers bonded to Si via alkylene bridges, (C) hydrosilylation catalysts, (D) ion-conductive compds. [e.g., alkali metal (or alkaline earth metal) salts, ionic liqs.], and (E) phenol-type antioxidants and satisfying molar ratio SiH/alkenyl 0.1-10:1, D/(A + B) 0.1-30 weight%, and E/(A + B) 0.1-10 weight%, and printer rolls having rubber layers of the above upon core metals are sep. claimed. The rolls exhibit less oil bleeding nor surface tack and stable elec. resistivity against voltage change.

IT 870680-33-4P

AB

(hydrosilylation monomers; ion-conductive silicone rubber rolls showing stable resistivity against voltage change for copiers)

RN 870680-33-4 HCAPLUS

CN Poly(oxy-1,2-ethanediy1), α-methy1-ω-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-y1)propoxy]- (9CI) (CA INDEX NAME)

IC ICM C08L083-07

ICS C08K003-00; C08K005-13; C08L083-04; G03G015-02; G03G015-08; G03G015-16; H01B001-20

C 74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

Section cross-reference(s): 39

T hydrosilylation curable silicone rubber printer roll; lithium trifluoromethanesulfonylimide ionic conductor electrophotog roller; tackiness suppressed printer roller silicone rubber; tocopherol antioxidant rubber roller resistivity stability

IT Electrophotographic apparatus

(rollers; ion-conductive silicone rubber rolls showing stable resistivity against voltage change for copiers)

IT 870680-33-4P

(hydrosilylation monomers; ion-conductive silicone rubber rolls showing stable resistivity against voltage change for copiers)

L22 ANSWER 5 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2005:1292076 HCAPLUS Full-text

DOCUMENT NUMBER: 144:38331

TITLE: Polysil

Polysiloxane-based compound and solid polymer electrolyte composition using the same

INVENTOR(S): Kang, Yongku; Lee, Changjin; Lee, Jun Kyoung
PATENT ASSIGNEE(S): Korea Research Institute of Chemical Technology,

SOURCE: SOURCE: U.S. Pat. Appl. Publ., 22 pp.

CODEN: USXXCO
DOCUMENT TYPE: Patent
LANGUAGE: English

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2005271948	A1	20051208	US 2005-146970	20050607
			<	
KR 2005116475	A	20051213	KR 2004-41502	20040607
			<	
JP 2005350673	A	20051222	JP 2005-167039	20050607
			<	
PRIORITY APPLN. INFO.:			KR 2004-41502 A	20040607
			<	

ED Entered STN: 09 Dec 2005

AB This invention relates to a polysiloxane-based compound and a solid polymer electrolyte composition prepared using the same. More particularly, the present invention relates to a polysiloxane-based polymer, which promotes easy crosslinking and also enables to control the level of crosslinking according to the concentration of an acryl group by introducing a polyalkyleneoxide group and an acryl group are introduced as side chains to the backbone of methyleiloxane polymer.

IT 362050-08-0P

(polysiloxane-based compound for solid polymer electrolyte composition)

RN 362060-08-0 HCAPLUS

CN Poly(oxy-1,2-ethanediy1), $\alpha,\alpha',\alpha'',\alpha'''$ -

[(2,4,6,8-tetramethylcyclotetrasiloxane-2,4,6,8-tetrayl)tetra-3,1-propanediyl]tetrakis[@-methoxy- (CA INDEX NAME)

$$\begin{array}{c} \text{MeO} & \begin{array}{c} \text{CH}_2 - \text{CH}_2 - \text{O} \\ \end{array} \end{array} \begin{array}{c} \text{N} \\ \text{Me} \\ \end{array} \begin{array}{c} \text{CH}_2 \end{array} \begin{array}{c} \text{Si} \\ \text{OCH}_2 \end{array} \begin{array}{c} \text{CH}_2 \end{array} \begin{array}{c} \text{CH}_2$$

PAGE 1-B

$$-CH_2$$
 OMe

$$--CH_2$$
 OMe

IT 870809-21-5P

(polysiloxane-based compound for solid polymer electrolyte composition)

RN 870809-21-5 HCAPLUS

CN 2-Propenoic acid, 2,4,6,8-tetramethylcyclotetrasiloxane-2,4,6,8-tetrayltetrakis(3,1-propanediyloxy-2,1-ethanediyl) ester (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

__o__C__cH___cH:

IC ICM H01M006-14

INCL 429302000

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

Section cross-reference(s): 38

ST battery electrolyte polysiloxane based compd

IT Polysiloxanes, uses

(polyoxyalkylene-; polysiloxane-based compound for solid polymer electrolyte composition)

IT Polyoxyalkylenes, uses

(polysiloxane-; polysiloxane-based compound for solid polymer

electrolyte composition)

T Battery electrolytes

Ionic conductivity

(polysiloxane-based compound for solid polymer electrolyte composition)

IT Polysiloxanes, uses

(polysiloxane-based compound for solid polymer electrolyte composition)

Plastics, uses

(thermoplastics; polysiloxane-based compound for solid polymer electrolyte composition)

IT 94-36-0, Benzoyl peroxide, processes

(polysiloxane-based compound for solid polymer electrolyte composition)

IT 33454-82-9, Lithium triflate

(polysiloxane-based compound for solid polymer electrolyte composition)

IT 26403-67-8DP, reaction products with allyl containing ethylene glycol derivs. 27252-80-8DP, reaction products with polymethylsiloxane 49718-23-2DP, reaction products with allyl containing ethylene glycol derivs. 70964-99-7DP, reaction products with polymethylsiloxane 362060-08-0P

(polysiloxane-based compound for solid polymer electrolyte composition)

IT 96-49-1, Ethylene carbonate 108-32-7, Propylene carbonate 24991-55-7, Polyethylene glycol dimethyl ether (polysiloxane-based compound for solid polymer electrolyte

composition)

IT 70964-99-7P 870809-21-5P

(polysiloxane-based compound for solid polymer electrolyte composition)

L22 ANSWER 6 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2005:369142 HCAPLUS Full-text DOCUMENT NUMBER: 142:431092

TITLE: Silicone composition for biocompatible membrane

INVENTOR(S): Tapsak, Mark A.; Valint, Paul PATENT ASSIGNEE(S): DexCom, Inc., USA

SOURCE: U.S. Pat. Appl. Publ., 34 pp.

CODEN: USXXCO DOCUMENT TYPE: Patent

LANGUAGE: English FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE ----US 2005090607 A1 20050428 US 2003-695636 20031028 <--A2 20050519 WO 2004-US35499 WO 2005045394 20041026 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC. VN. YU. ZA. ZM. ZW RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW,

DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG US 2008045824 A1 20080221 US 2007-763215 20070614

PRIORITY APPLN. INFO .:

Entered STN: 29 Apr 2005

US 2003-695636 A 20031028 <---

ED AB A biocompatible membrane comprises a silicone composition comprising a

hydrophile covalently incorporated therein, wherein the biocompatible membrane controls transport of an analyte through the membrane. The polymeric material can be useful as a biocompatible membrane for use in biosensor applications.

- 362060-08-0P 850714-45-3P
- (silicone composition for biocompatible membrane)
- 362060-08-0 HCAPLUS RN

CN Poly(oxy-1,2-ethanediyl), $\alpha,\alpha',\alpha'',\alpha'''$ -[(2,4,6,8-tetramethylcvclotetrasiloxane-2,4,6,8-tetrayl)tetra-3,1propanediyl]tetrakis[@-methoxy- (CA INDEX NAME)

$$\begin{array}{c} \text{PAGE } 1-\text{A} \\ \text{MeO} & \begin{array}{c} \text{CH}_2 - \text{CH}_2 - \text{O} \\ \end{array} \end{array} \begin{array}{c} \text{Ne} \\ \text{Me} \\ \end{array} \begin{array}{c} \text{Me} \\ \text{CH}_2 - \text{CH}_2 - \text{O} \\ \end{array} \begin{array}{c} \text{Ne} \\ \text{CH}_2 \end{array} \begin{array}{c} \text{Si} \\ \text{Me} \\ \end{array} \begin{array}{c} \text{CH}_2 \end{array} \begin{array}{c} \text{Si} \\ \text{Me} \\ \end{array} \begin{array}{c} \text{CH}_2 \end{array} \begin{array}{c} \text{Si} \\ \text{CH}_2 \end{array} \begin{array}{c} \text{CH}_2 \end{array} \begin{array}{c} \text{CH}_2 \\ \text{CH}_2 \end{array} \begin{array}{c} \text{CH}_2 \end{array} \begin{array}{c} \text{CH}_2 \\ \text{CH}_2 \end{array} \begin{array}{c} \text{CH}_2 \end{array} \begin{array}{c} \text{CH}_2 \\ \text{CH}_2 \end{array}$$

PAGE 1-B

- 850714-45-3 HCAPLUS RN
- CN Cyclotetrasiloxane, octamethyl-, polymer with α - $(\texttt{ethenyldimethylsilyl}) - \omega - \texttt{[(ethenyldimethylsilyl)oxy]} \\ \texttt{poly[oxy(dimethylsilyl)oxy]} \\$ ethylsilvlene)] and $\alpha.\alpha'.\alpha''.\alpha'''-[(2.4.6.8$ tetramethylcyclotetrasiloxane-2, 4, 6, 8-tetrayl)tetra-3, 1propanediyl]tetrakis[ω-methoxypoly(oxy-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 362060-08-0

CMF (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n C20 H48 O8 Si4

CCI PMS

PAGE 1-B

CM 2

CRN 59942-04-0

CMF (C2 H6 O Si)n C8 H18 O Si2

CCI PMS

CM 3

CRN 556-67-2

CMF C8 H24 O4 Si4

ICM C08L083-00

INCL 524588000

CORPORATE SOURCE:

ED

ΙT

CC 37-6 (Plastics Manufacture and Processing)

ΙT 362060-08-0P 850714-45-3P

(silicone composition for biocompatible membrane)

L22 ANSWER 7 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2004:938465 HCAPLUS Full-text

DOCUMENT NUMBER: 142:94499

TITLE: The polymer electrolyte based on polysiloxane containing both alkyl cyanide and

oligo ethylene oxide pendants Lee, Young Sik; Song, Gi Sang; Kang, Yongku; Suh, AUTHOR(S):

Dong Hack

Advanced Materials Division, Korea Research Institution of Chemical Technology, Yusong Taejon,

305-600, S. Korea

SOURCE: Electrochimica Acta (2004), 50(2-3), 311-316

CODEN: ELCAAV; ISSN: 0013-4686

PUBLISHER: Elsevier B.V. DOCUMENT TYPE: Journal LANGUAGE: English Entered STN: 08 Nov 2004

AB methoxyethoxy)ethoxy)propyl) methylsiloxane cyclics (D4TEG) were prepared by a hydrosilvlation reaction of 2,4,6,8tetramethylcyclotetrasiloxane (D4H) with allyl cyanide and tri(ethylene glycol) Me allyl ether (allyl TEG), resp., in toluene using a platinum(0)-1,3divinyl-1,1,3,3-tetramethyl disiloxane complex as the catalyst. The new crosslinkable polymers with alkyl cyanide and ethylene oxide groups as the pendent were synthesized by ring opening polymerization and characterized by GPC and 1H NMR. And then, the crosslinked solid polymer electrolyte was prepared by UV radiation curing. The conductivities of samples were measured by impedance spectroscopy using an indium-tin oxide (ITO) electrode. As the results, the maximum ionic conductivities of the polymer were 1.15 + 10-5 S

cm-1 at 20 °C and 1 + 10-4 S cm-1 at 60 °C. The electrolyte was stable

electrochem. to 5 V vs. Li+/Li at room temperature

819800-12-9DP, reaction products with tetravinyldimethyldisiloxane, lithium complexes

(polymer *lectrolyte based on polysiloxane containing both alkyl cyanide and oligo ethylene oxide pendants)

RN 819800-12-9 HCAPLUS

CN Cyclotetrasiloxane-2,4,6,8-tetrabutanenitrile, 2,4,6,8-tetramethyl-, polymer with 2,4,6,8-tetramethyl-2,4,6,8-tetrakis(4,7,10,13tetraoxatetradec-1-y1)cyclotetrasiloxane (9CI) (CA INDEX NAME)

CM 1

CRN 131718-86-0 CMF C44 H96 O20 Si4

PAGE 1-A

(CH2) 3 - O

Me

MeO-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2-O-CH2) 3

Me

MeO-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-C

PAGE 1-B

$$-$$
 (CH2) 3-0-CH2-CH2-O-CH2-CH2-O-CH2-OH2-OMe

CM 2

CRN 1448-49-3

CMF C20 H36 N4 O4 S14

- CC 37-3 (Plastics Manufacture and Processing)
- ST polymer electrolyte polysiloxane cyanide oligooxyethylene
- lithium; ionic cond polysiloxane cyanide oligooxyethylene lithium
- T Polysiloxanes, preparation
 - (lithium complexes; polymer electrolyte based on polysiloxane containing both alkyl cyanide and oligo ethylene oxide pendants)
- IT Electric impedance Ionic conductivity Polymer electrolytes

(polymer electrolyte based on polysiloxane containing both alkyl cyanide and oligo ethylene oxide pendants) 7439-93-2DP, Lithium, complexes with polysiloxanes reaction products with polysiloxanes, lithium complexes 819800-12-90P, reaction products with tetravinyldimethyldisiloxane, lithium complexes (polymer electrolyte based on polysiloxane containing both

alkyl cyanide and oligo ethylene oxide pendants)

REFERENCE COUNT: 29 THERE ARE 29 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L22 ANSWER 8 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN 2004:782571 HCAPLUS Full-text ACCESSION NUMBER.

DOCUMENT NUMBER: 143:26659

TITLE: Regioselective hydrosilvlation of allylic

derivatives by the siloxanes D4H and MDHM in the presence of platinum catalysts

AUTHOR(S): El Malki, A.; Hannioui, A.; Knouzi, N.; Vaultier,

CORPORATE SOURCE: Universite Cadi Ayyad. FST Beni-Mellal, Beni-Mellal, Morocco

SOURCE: Physical & Chemical News (2004), 17(1),

107-112 CODEN: PCNHBU; ISSN: 1114-3800

Best Edition Journal

LANGUAGE: English CASREACT 143:26659 OTHER SOURCE(S):

ED Entered STN: 27 Sep 2004

New functionalized cyclic and acyclic siloxanes have been prepared in high AB vields by the platinum catalyzed hydrosilylation of terminal alkenes functionalized in the allylic positions. Conditions have been found that allow the regioselective obtention of siloxanes resulting exclusively from a β-addition of D4H or MDHM to the double bond. This regionelectivity depend on several factors including solvent, catalyst, amount of catalyst, reaction temperature and order of addition of reactants. The optimized conditions are compatible with the presence of free alc. groups thus allowing the direct preparation of functionalized siloxanes. All new compds. were characterized by 1H, 13C, 29Si NMR and HRMS as well.

IT 852954-43-9P

PUBLISHER:

DOCUMENT TYPE:

(regioselective hydrosilylation of allylic derivs. by hydro

siloxanes in presence of platinum catalysts)

852954-43-9 HCAPLUS RN

2-Propanol, 1,1',1'',1'''-[(2,4,6,8-tetramethylcyclotetrasiloxane-CN 2,4,6,8-tetrayl)tetrakis[3,1-propanediyloxy(1-methyl-2,1ethanediyl)oxy]]tetrakis- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

DOCUMENT NUMBER:

29-6 (Organometallic and Organometalloidal Compounds) 7422-52-8P 20995-84-0P 60665-85-2P 849830-58-6P 852954-40-6P 852954-41-7P 852954-42-8P 852954-43-9P 852954-44-0P

(regioselective hydrosilylation of allylic derivs, by hydro

siloxanes in presence of platinum catalysts)

THERE ARE 16 CITED REFERENCES AVAILABLE FOR REFERENCE COUNT: 16 THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L22 ANSWER 9 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2004:670318 HCAPLUS Full-text

141:424716 TITLE: Synthesis and conductivity study on

oligoethyleneoxy-functionalized cyclic siloxanes AUTHOR(S): Zhang, Zhengcheng; Simon, Ann; Jin, Jay J.; Lyons,

Leslie J.; Amine, Khalil; West, Robert

CORPORATE SOURCE: Organosilicon Research Center, Department of Chemistry, University of Wisconsin-Madison,

Madison, WI, 53706, USA

SOURCE: PMSE Preprints (2004), 91, 587-588 CODEN: PPMRA9; ISSN: 1550-6703 PUBLISHER: American Chemical Society DOCUMENT TYPE: Journal; (computer optical disk)

LANGUAGE: English ED Entered STN: 18 Aug 2004

Pentamethylcyclopentasiloxane (D5H) comb polymers 1 and 2 with oligoethyleneoxy substituents were synthesized by dehydrogenative silation and

hydroslilylation reaction. The formation of M and T species was observed in

BICGF5)3 catalyzed decoupling reaction and the mechanism was proposed. Ionic conductivities of these polymers complexes were measured by a.c. impedance method after doping with LiN(CF3SO2)2. The oxygen-linked cyclic siloxane derivative 1 exhibits higher conductivity than the trimethylene-linked 2 at the same LiTFSI doping level, probably due to its lower viscosity. The linear short-chain siloxane counterpart 3, investigated in our earlier study, has a conductivity (2.7x10-45/cm) almost twice as large as cyclic siloxane 1 at 25°.

IT 795296-82-1P

(synthesis and conductivity of oligoethyleneoxy-functionalized cyclic ${\tt siloxanes}$)

RN 795296-82-1 HCAPLUS

CN Cyclopentasiloxane, 2,4,6,8,10-pentamethyl-2,4,6,8,10-pentakis(4,7,10,13-tetraoxatetradec-1-yl)- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

PAGE 1-C

--- CH2-- OMe

CC 37-3 (Plastics Manufacture and Processing)

IT 795296-82-1P

(synthesis and conductivity of oligoethyleneoxy-functionalized cyclic siloxanes)
REFERENCE COUNT: 23 THERE ARE 23 CITED REFERENCES AVAILABLE FOR

THIS RECORD. ALL CITATIONS AVAILABLE IN THE

RE FORMAT

L22 ANSWER 10 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:392049 HCAPLUS Full-text DOCUMENT NUMBER: 140:395247

TITLE: Cosmetic hydrophilized powder surface treated with

polyether-modified silicone INVENTOR(S): Kamei, Masanao; Tachibana, Kivomi

PATENT ASSIGNEE(S): Shin-Etsu Chemical Co., Ltd., Japan

SOURCE: U.S. Pat. Appl. Publ., 17 pp. CODEN: USXXCO

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2004091440	A1	20040513	US 2003-701566	20031106
US 7285290	B2	20071023	<	
JP 2004155978	A	20040603	JP 2002-324840	20021108
JP 3979922	B2	20070919	<	
EP 1424373	A2	20070919	EP 2003-78528	20031110
			<	

EP 1424373 A3 20050525

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK PRIORITY APPLN. INFO .: JP 2002-324840 A 20021108 <--

ED Entered STN: 14 May 2004

AB The present invention is hydrophilized powder, wherein the powder is surface treated with polyether-modified silicone having a hydrolyzable silyl group. The invention also provides a composition comprising the powder, an aqueous dispersion comprising the powder, and their application in cosmetics, coatings, and inks. Me3SiO(SiMe2O)10(SiRMeO)3(SiR1Me)2S iMe3 [R = C3H6O(CH2CH2O)32Me, R1 = C2H4Si(OEt)3] was prepared and used in a composition also containing TiO2. A number of cosmetic examples including sunscreens, eyeshadow, mascara, etc. containing the siloxanes were given.

686342-81-4P

(cosmetic hydrophilized powder surface treated with

polyether-modified silicone)

686342-81-4 HCAPLUS RN

CN Poly(oxy-1,2-ethanediyl), α,α' -[[6,8-bis[2-

(triethoxysily1)ethy1]-2,4,6,8-tetramethylcyclotetrasiloxane-2,4-

diyl]di-3,1-propanediyl]bis[ω-methoxy- (9CI) (CA INDEX NAME)

PAGE 1-B

C ICM A61K007-06

ICS A61K007-11; A61K007-021

INCL 424070120; X42-4 6.3

CC 62-4 (Essential Oils and Cosmetics)

Section cross-reference(s): 42

IT 686342-81-4P

(cosmetic hydrophilized powder surface treated with

polyether-modified silicone)

REFERENCE COUNT: 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR

THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L22 ANSWER 11 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2003:551208 HCAPLUS Full-text DOCUMENT NUMBER: 139:101535

TITLE:

Production of oxyalkylene-containing

acrylate-terminated polysiloxane crosslinking agents

INVENTOR(S): Kang, Yongku; Lee, Changjin; Lee, Won Sil; Noh, Kun Ae

PATENT ASSIGNEE(S): Korea Research Institute of Chemical Technology, S. Korea

SOURCE: U.S. Pat. Appl. Publ., 18 pp.

CODEN: USXXCO
DOCUMENT TYPE: Patent

LANGUAGE: English FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2003134968	A1	20030717	US 2002-282214	20021028
US 6783897	B2	20040831	•	0004444
KR 2003040618	A	20030523	KR 2001-70969 <	20011115
JP 2003277506	A	20031002	JP 2002-324866 <	20021108
JP 3749217 PRIORITY APPLN. INFO.:	B2	20060222	KR 2001-70969 A	20011115
INIONIII ALPEN. INFO.:			KR 2001-70909 A	20011113

ED Entered STN: 18 Jul 2003

- A crosslinking agent comprises Me siloxane polymer backbone, a poly(alkylene AB oxide) branches and from 2 to 4 of acrylate groups at both terminals. A solid polymer electrolyte composition comprises (a) 0.1-80% of the crosslinking agent, (b) 0.1-80% of a plasticizer selected from poly(alkylene glycol) dialkyl ethers and non-aqueous polar solvents, (c) 3-30% of a lithium salt, and (d) 0.5-5% of a curing initiator. The crosslinkable solid polymer edectrolyte composition has a high ionic conductivity at room temperature and can be readily formed into a film suitable for use in large-size lithiumpolymer secondary batteries applicable to elec. cars, power storage devices for power leveling, as well as in small-size lithium-polymer secondary batteries applicable to video cameras and portable data terminals, such as cellular phones and notebook computers. Thus, tri(ethylene glycol) allyl Me ether was hydrosilylated with 2,4,6,8-tetramethylcyclotetrasiloxane in the presence of a platinum catalyst producing tetrafunctional tri(ethylene glycol)-substituted D4 monomer in 97.4% yield. The monomer was polymerized in the presence of 1,3-di(3-acryloyloxypropyl)-1,1,3,3- tetramethyldisiloxane terminating agent and sulfuric acid to obtain a polyoxyethylene-grafted acryloyloxy-terminated polysiloxane used as a crosslinkable component in solid polymer electrolyte compns.
- IT 131718-86-0P 362060-08-0P

(monomer; production of oxyalkylene-containing acrylate-terminated polysiloxane crosslinking agents)

- RN 131718-86-0 HCAPLUS
- CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-tetrakis(4,7,10,13-tetraoxatetradec-1-vl)- (9CI) (CA INDEX NAME)

- $-- (\mathtt{CH}_2) \ \mathtt{3} \mathtt{O} \mathtt{CH}_2 \mathtt{CH}_2 \mathtt{O} \mathtt{CH}_2 \mathtt{CH}_2 \mathtt{O} \mathtt{CH}_2 \mathtt{CH}_2 \mathtt{O} + \mathtt{CH}_2 \mathtt{CH}_2 \mathtt{O} + \mathtt{CH}_2 \mathtt{CH}_2$
- RN 362060-08-0 HCAPLUS
- CN Poly(oxy-1,2-ethanediy1), a,a',a'',a'''-[(2,4,6,8-tetramethylcyclotetrasiloxane-2,4,6,8-tetrayl)tetra-3,1propanediyl]tetrakis(o-methoxy- (CA INDEX NAME)

PAGE 1-B

____CH2_______OMe

- IT 561065-50-7DP, acryloyloxy-terminated 561065-52-9DP, acryloyloxy-terminated 561065-55-2DP, acryloyloxy-terminated (production of oxyalkylene-containing acrylate-terminated polysiloxane crosslinking agents)
- RN 561065-50-7 HCAPLUS
- CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-tetrakis(4,7,10,13-tetraoxatetradec-1-yl)-, homopolymer (9CI) (CA INDEX NAME)

CM

CRN 131718-86-0 CMF C44 H96 O20 Si4

PAGE 1-B

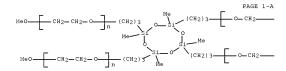
- RN 561065-52-9 HCAPLUS
- CN Poly(oxy-1,2-ethanediyl), $\alpha,\alpha',\alpha'',\alpha'''$ -[(2,4,6,8-tetramethylcvclotetrasiloxane-2,4,6,8-tetrayl)tetra-3,1propanediyl]tetrakis[ω-methoxy-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 362060-08-0

CMF (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n C20 H48 O8 Si4

CCI PMS



PAGE 1-B

- RN 561065-55-2 HCAPLUS
- CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-tetrakis(4,7,10,13tetraoxatetradec-1-yl)-, polymer with α,α',α'' , alph a.'''-[(2,4,6,8-tetramethylcyclotetrasiloxane-2,4,6,8-tetrayl)tetra-3,1-propanediyl]tetrakis[ω-methoxypoly(oxy-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 362060-08-0

CMF (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n C20 H48 O8 Si4 CCI PMS

$$\begin{array}{c} \text{PAGE 1-A} \\ \text{MeO} & \begin{array}{c} \text{CH}_2\text{-}\text{CH}_2\text{-}\text{O} \\ \end{array} \\ \text{MeO} & \begin{array}{c} \text{CH}_2\text{-}\text{O} \\ \end{array} \\ \text{CH}_2\text{-}\text{CH}_2\text{-}\text{O} \\ \end{array} \\ \text{MeO} & \begin{array}{c} \text{CH}_2\text{-}\text{O} \\ \end{array} \\ \text$$

PAGE 1-B

CM 2

CRN 131718-86-0 CMF C44 H96 O20 Si4

PAGE 1-B

-- (CH2) 3-0-CH2-CH2-O-CH2-CH2-O-CH2-OH2-OMe

IC ICM C08F008-00

INCL 524588000

35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 52

acrylate terminated polyoxyalkylene graft polysiloxane crosslinking agent; lithium salt acrylate terminated polyoxyalkylene polysiloxane solid electrolyte; solid polymer ionic conductor lithium secondary battery

ΙT Plastic films

(from solid electrolytes based on lithium salts and

oxyalkylene-containing acrylate-terminated polysiloxanes)

Ionic conductors

Solid electrolytes

(solid electrolytes based on lithium salts and

oxvalkylene-containing acrylate-terminated polysiloxanes)

131718-86-0P 362060-08-0P 561065-47-2P

561065-48-3P

(monomer; production of oxyalkylene-containing acrylate-terminated

polysiloxane crosslinking agents)

561065-50-7DP, acryloyloxy-terminated 561065-51-8P

561065-52-9DP, acryloyloxy-terminated 561065-53-0P

561065-55-2DP, acryloyloxy-terminated (production of oxyalkylene-containing acrylate-terminated polysiloxane

crosslinking agents)

REFERENCE COUNT: 18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE

RE FORMAT

L22 ANSWER 12 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN 2002:792190 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 137:311716

TITLE: Polyalkylene oxide porogens having hyper-branches and low dielectric-constant insulators using them Lee, Changjin; Kong, Yongku; Kang, Yong Goo; Kim, INVENTOR(S):

Hee Jungo; Jin, Moon Young; Seok, Sang Ii; Char, Kookheon; Chu, Sang-Hyun

PATENT ASSIGNEE(S): Korea Research Institute of Chemical Technology, S. Korea

SOURCE: Eur. Pat. Appl., 34 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PA:	TENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP	1249846	A2	20021016	EP 2002-7257	20020328
				<	
EP	1249846	A3	20021127		
	R: AT, BE, CH,	DE, DK	, ES, FR, GB	, GR, IT, LI, LU, NL,	SE, MC,
	PT, IE, SI,	LT, LV	, FI, RO, MK	, CY, AL, TR	
KR	2002080196	A	20021023	KR 2001-19623	20010412
				<	
US	2003078443	A1	20030424	US 2002-118094	20020409
				<	
US	6914147	B2	20050705		
JP	2003040998	A	20030213	JP 2002-110253	20020412

PRIORITY APPLN. INFO.:

KR 2001-19623 A 20010412

ED Entered STN: 18 Oct 2002

AB Low dielec.—constant insulators having nanopores are prepared by coating a mixture a polyalkylene oxide porogen and a heat-resistant resin (e.g., a silsesquioxane) on a substrate and thermally treating the coated substrate at a temperature effective to degrade the porogen. A porogen was prepared by esterification of dipentaerythritol with methoxypolyethyleneoxy acetic acid.

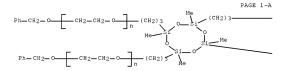
IT 470722-04-4P

(polyalkylene oxide porogens having hyper-branches and low dielec.-constant insulators using them)

RN 470722-04-4 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), $\alpha,\alpha',\alpha'',\alpha'''$ -

[(2,4,6,8-tetramethylcyclotetrasiloxane-2,4,6,8-tetrayl)tetra-3,1-propanediyl]tetrakis[ω-(phenylmethoxy)- (9CI) (CA INDEX NAME)



PAGE 1-B

IC ICM H01B003-46

ICS C09D105-00; C09D171-02; C09D183-04; C09D201-00; H01B003-18; C08J009-00; C08J009-16; C08J009-228

CC 37-6 (Plastics Manufacture and Processing)

Section cross-reference(s): 76

-- access to the state of the

IT 86321-17-7P 470722-04-4P

(polyalkylene oxide porogens having hyper-branches and low dielec.-constant insulators using them)

L22 ANSWER 13 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2002:658762 HCAPLUS Full-text

DOCUMENT NUMBER: 137:186376

TITLE: Hydrophilic curable ethoxylated silicones

INVENTOR(S): Gosselink, Eugene Paul; Trinh, Toan; Gardner, Robb

Richard

PATENT ASSIGNEE(S): The Procter & Gamble Co., USA SOURCE: U.S. Pat. Appl. Publ., 22 pp.

U.S. Pat. Appl. Publ., 22 pp. CODEN: USXXCO

DOCUMENT TYPE: Patent
LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PATENT NO.				KIND DATE			APPLICATION NO.								
	2002						2002	0829			001-					0011115
					B2 20031118 A1 20030109			1	WO 2			690		2	0011114	
										<						
	W:	CN,	co,	CR,	CU,	CZ,	AU, DE, ID,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,
		LC,	LK, NZ,	LR, PH,	LS, PL,	LT,	LU, RO,	LV, RU,	MA, SD,	MD, SE,	MG, SG,	MK,	MN,	MW,	MX,	MZ,
	RW:	GH,	GM,	KE,	LS,	MW,	UZ,	SD,	SL,	SZ,	TZ,					
		SE,		BF,	ВJ,		FI, CG,									
A	J 2002						2003	0303	i	AU 2		2366 	05		2	0011114
E	2 1334	1229			A1		2003	0813	1	EP 2		9861 	39		2	0011114
	R:						ES,						LU,	NL,	SE,	MC,
JE	2004										003-		68		2	0011114
PRIORI	TY APE	LN.	INFO	. :					1	US 2		2492	34P		P 2	0001116
									1	WO 2	001-		690	,	W 2	0011114

ED Entered STN: 30 Aug 2002

BH Hydrophilic curable alkoxylated silicone polymers (for surface modification) are useful in, e.g., fiber and fabric care, hair care, skin care, surface care, and car care compns. The compds. are curable silicone polymers which contain 21 polyalkyleneoxy groups, preferably polyalkyleneoxy pendant groups, comprising at least some ethyleneoxy units, the polyalkyleneoxy pendant groups are preferably capped with low mol. weight alkyl groups, such as Cl-6-alkyl. These compds. are substantive to the surface but keep the surface hydrophilic. A polyethylene glycol allyl Me ether intermediate was reacted with Me terminated methylhydrosiloxane-dimethylsiloxane copolymer in the presence of Pt catalyst, subsequently M-allylethylenediamine and vinylmethyldimethoxysilane to give curable silicone with amine, Me ethoxylate, and SiOMe functionality.

IT 362060-08-0P

⁽intermediate for adduction with reactive siloxanes; hydrophilic curable ethoxylated silicones and intermediate functional ethoxylates)

RN 362060-08-0 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), $\alpha,\alpha',\alpha'',\alpha'''$ -

^{[(2,4,6,8-}tetramethylcyclotetrasiloxane-2,4,6,8-tetray1)tetra-3,1-propanediyl]tetrakis[ω -methoxy- (CA INDEX NAME)

$$\begin{array}{c} \text{PAGE } 1-\text{A} \\ \text{MeO} & \begin{array}{c} \text{CH}_2-\text{CH}_2-\text{O} \end{array} \end{array} \\ \begin{array}{c} \text{Me} \\ \text{Me} \end{array} \\ \begin{array}{c} \text{Me} \\ \text{Si} \end{array} \\ \begin{array}{c} \text{CH}_2 \end{array}) \\ \begin{array}{c} \text{Si} \\ \text{CH}_2 \end{array}) \\ \begin{array}{c} \text{Me} \\ \text{CH}_2 \end{array} \\ \begin{array}{c} \text{CH}_2 \end{array}) \\ \begin{array}{c} \text{Si} \\ \text{Me} \end{array} \\ \end{array} \\ \begin{array}{c} \text{CH}_2 \end{array}) \\ \begin{array}{c} \text{Si} \\ \text{Me} \end{array} \\ \begin{array}{c} \text{CH}_2 \end{array}) \\ \begin{array}{c} \text{Si} \\ \text{Me} \end{array} \\ \end{array} \\ \begin{array}{c} \text{CH}_2 \end{array}) \\ \begin{array}{c} \text{Si} \\ \text{Me} \end{array} \\ \begin{array}{c} \text{CH}_2 \end{array}) \\ \begin{array}{c} \text{Si} \\ \text{Me} \end{array} \\ \end{array} \\ \begin{array}{c} \text{CH}_2 \end{array}) \\ \begin{array}{c} \text{Si} \\ \text{Me} \end{array} \\ \begin{array}{c} \text{CH}_2 \end{array}) \\ \begin{array}{c} \text{Si} \\ \text{Me} \end{array} \\ \begin{array}{c} \text{CH}_2 \end{array}) \\ \begin{array}{c} \text{Si} \\ \text{Me} \end{array} \\ \end{array} \\ \begin{array}{c} \text{CH}_2 \end{array} \\ \\ \begin{array}{c} \text{CH}_2 \end{array} \\ \begin{array}{c} \text{CH}_2$$

PAGE 1-B

TC ICM C08J003-00

INCL 524588000

37-3 (Plastics Manufacture and Processing)

ΙT 9041-33-2P, Ethylene oxide-propylene oxide copolymer monoallyl ether 27252-80-8P, Polyethylene glycol allyl methyl ether 27274-31-3P, Polyethylene glycol monoallyl ether 52002-43-4P 97969-60-3P 147962-80-9P 189240-06-0P 362060-08-0P 449754-21-6P 449754-22-7P 449754-23-8P 449754-24-9P 449754-25-0P 449754-26-1P

(intermediate for adduction with reactive siloxanes; hydrophilic curable ethoxylated silicones and intermediate functional ethoxvlates)

L22 ANSWER 14 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2002:593249 HCAPLUS Full-text DOCUMENT NUMBER: 137:311842

TITLE: Characteristics of low-k nanoporous poly(methylsilsequioxane) thin films

AUTHOR(S): Min, Sung-Kvu; Park, Jae-Man; Song, Kitae; Jin,

Moon Young; Lee, Changjin; Yoon, Do Yeung; Rhee, Hee-Woo

CORPORATE SOURCE: Dept. of Chemical Engineering, Sogang University, Seoul, 121-742, S. Korea

SOURCE: Molecular Crystals and Liquid Crystals Science and Technology, Section A: Molecular Crystals and

Liquid Crystals (2002), 377, 193-196

CODEN: MCLCE9: ISSN: 1058-725X

PUBLISHER: Taylor & Francis Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

Entered STN: 09 Aug 2002

AB Nanoporous poly(Me silsesquioxane) (PMSSQ) was obtained by sintering organic/inorg, nanohybrids. The porogen was cyclosiloxane with four poly(ethylene glycol) arms for better miscibility with PMSSQ. As the porogen content in the hybrid increased up to 30 vol%, the porosity of the calcined PMSSQ film increased up to 24%, and the k values decreased as low as 2.12. However, the interconnected pore structure was observed at the porogen content above 25 vol%. The miscibility was improved compared to poly(caprolactone)-based porogens and the pore size was indistinguishable even at SEM resolution 13 36206-06-0

(characteristics of low-k nanoporous poly(methylsilsequioxane) dielec. thin films obtained with cyclosiloxane derivative having four polyethylene glycol arms as porogen)

RN 362060-08-0 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α,α',α'',α''', α''', (2,4,6,8-tetranethylcyclotetrasiloxane-2,4,6,8-tetrayl)tetra-3,1-propanediyl)tetrakis(ω-methoxy- (CA INDEX NAME)

PAGE 1-B

$$-CH_2$$
 oMe

- CC 38-3 (Plastics Fabrication and Uses) Section cross-reference(s): 37
- IT 362060-08-0

(characteristics of low-k nanoporous poly(methylsilsequioxane) dielec. thin films obtained with cyclosiloxane derivative having four polyethylene qlycol arms as porogen)

REFERENCE COUNT: 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L22 ANSWER 15 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NOMBER: 2002:391836 HCAPLUS Full-text DOCUMENT NUMBER: 136:403510

TITLE: Fabric color care method for spray application to

garments INVENTOR(S):

Trinh, Toan; Barnabas, Mary Vijayarani; Gosselink, Eugene Paul; Smith, John William; Tordil, Helen Bernardo; Gardner, Robb Richard; Coffindaffer,

Timothy Woodrow

PATENT ASSIGNEE(S): The Procter & Gamble Company, USA SOURCE:

PCT Int. Appl., 87 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Pat.ent. LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

									APPLICATION NO.								
												001-				20011114	
		W:	CN, GE, LC, NO,	CO, GH, LK, NZ,	CR, GM, LR, PH,	CU, HR, LS, PL,	CZ, HU, LT, PT,	DE, ID, LU, RO,	DK, IL, LV, RU,	DM, IN, MA, SD,	DZ, IS, MD, SE,	BG, EC, JP, MG, SG,	BR, EE, KE, MK,	ES, KG, MN,	FI, KP, MW,	GB KR MX	, CH, , GD, , KZ, , MZ,
			GH, CY, TR,	GM, DE, BF,	KE, DK, BJ,	LS, ES, CF,	MW, FI, CG,	FR, CI,	SD, GB, CM,	SL, GR, GA,	SZ, IE, GN,	TZ, IT, GQ,	LU, GW,	MC, ML,	NL, MR,	PT NE	, CH, , SE, , SN,
	CA	2424	698			Al		2002	0523		CA 2		2424	698			20011114
	AU	2002	0269	16		A		2002	0527		AU 2	002-	2691	6			20011114
	EP	1341	892			A1		2003	0910		EP 2	001-	 9958 	66			20011114
	EP	1341 R:	AT,	BE,	CH,	DE,	DK,		FR,	GB,				LU,	NL,	SE	, MC,
	ΑT	3310										001-	9958	66			20011114
	US	2002	1122	93		A1		2002	0822		US 2	001-	 9992 	70			20011115
	US US	6790 2004	819 2213	97		B2 A1		2004 2004	0914 1111		US 2	004-		89			20040609
PRIO		7005 APP				В2		2006	0228		US 2		2492	42P		P	20001116
											WO 2		US43			W	20011114
											US 2	001-				A1	20011115

ED Entered STN: 24 May 2002

AB Methods for restoring and/or rejuvenating color of worn, faded fabric, comprise applying to the fabric a fabric color care composition of watersoluble and/or water-dispersible cationic polymers, reactive silicone polymers or aminosilicone polymers, and their mixts., surfactant capable of forming a bilayer structure, and mixts., optionally, other ingredients to provide addnl. fabric care benefits, and/or to improve performance and formulatability, as small particle size droplets, especially from a spray container. An example spray emulsion contained GE 176-12669 1.43, SM 2658 1.43, poly(vinyl alc.) 0.065, glycerin 0.01, Kathon CG 3 ppm, perfume and distilled water.

- IT 362060-08-3DP, tetrahydropyranyl-terminated, (hydrolyzed)
 (fabric color care method for spray application to garments)
- RN 362060-08-0 HCAPLUS
- CN Poly(oxy-1,2-ethanediy1), α,α',α'',α''', ((2,4,6,8-tetramethylcyclotetrasiloxane-2,4,6,8-tetray1)tetra-3,1-propanediv1ltetrakis(@-methoxy- (CA INDEX NAME)

PAGE 1-B

- IC ICM C11D003-37
- CC 46-5 (Surface Active Agents and Detergents)
- Section cross-reference(s): 40
- 2155-94-4DP, Dimethylallylamine, reaction products with Me H siloxane, and vinylmethyldiacetoxysilane 2944-70-9DP, reaction products with Me H siloxane, and dimethylallylamine 16753-62-1DP, Vinvlmethyldimethoxysilane, reaction products with Me H siloxane copolymers 27252-80-8DP, Polyethylene glycol allyl methyl ether, reaction products with Me H siloxane, and N-allylethylenediamine 40510-22-3DP, N-Allylethylenediamine, reaction products with polyethylene glycol allyl Me ether and Me H siloxane 52232-27-6DP, Ethylene oxide-propylene oxide copolymer allyl methyl ether, reaction products with Me H siloxane, and N-allylethylenediamine 85191-11-3DP, ethoxylated 147962-80-9DP, reaction products with Me H siloxane, and vinylmethyldimethoxysilane 156118-35-3DP, Methylsilanediol-dimethylsilanediol copolymer, Me-terminated, reaction products with N-allylethylenediamine and polyethylene glycol allyl Me ether 362060-08-0DP, tetrahydropyranyl-terminated, (hydrolyzed) 429689-29-2DP, reaction products with Me H siloxane, and vinvlmethyldimethoxysilane

(fabric color care method for spray application to garments)
REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR
THIS RECORD. ALL CITATIONS AVAILABLE IN THE
RE FORMAT

L22 ANSWER 16 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2001:703791 HCAPLUS Full-text

DOCUMENT NUMBER: 135:257630

TITLE: Organosilicon compositions from cyclosiloxanes
INVENTOR(S): Ferritto, Michael Salvatore; Schulz, William

James, Jr.

PATENT ASSIGNEE(S): Dow Corning Corporation, USA

SOURCE: U.S., 8 pp.

SOURCE: U.S., 8 pp. CODEN: USXXAM

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PAT	ENT NO.	KI	ND DATE	API	PLICATION NO.		DATE
US	6294634	В	1 2001	0925 US	2000-536141		20000328
EP	1138715	A	1 2001	1004 EP	2001-301491		20010220
		BE, CH, DE			R, IT, LI, LU,	NL, SE	, MC,
JP	200127898				2001-87220		20010326
PRIORITY	APPLN. I	INFO.:		US	2000-536141	A	20000328

ED Entered STN: 26 Sep 2001

ED Entered SIR: 26 Sep 2001

A Organosilicon compns. are prepared by heating various mixts. of (i) dimethylcyclosiloxanes or methylhydrogencyclosiloxanes, (ii) homopolymeric and copolymeric cyclosiloxanes containing 2C5 group, and (iii) homopolymeric and copolymeric cyclosiloxanes containing an oxyalkylene segment, in the presence of a ring opening catalyst e.g. hydroxides, at 30-250°. Thus, 2.1 g of 50 cSt polydimethylsiloxane fluid, 31.9 g of octamethylcyclotetrasiloxane, and 11.0 g of 1,3,5,7-tetramethyl-1,3,5,7-tetra(1-octyl)cyclotetrasiloxane were stirred in the presence of trifluoromethanesulfonic acid and heated to 75° for 4 h to give a linear siloxane.

IT 362060-08-0P 362060-09-1P 362060-10-4P

(polymerization of cyclosiloxanes, (co)polymer cyclosiloxanes, and (co)polymer cyclosiloxane-oxyalkylenes)

RN 362060-08-0 HCAPLUS

CN Poly(oxy-1,2-ethanedivl), $\alpha,\alpha',\alpha'',\alpha'''$ -

[(2,4,6,8-tetramethylcyclotetrasiloxane-2,4,6,8-tetrayl)tetra-3,1-

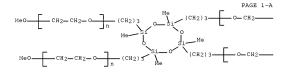
propagedivl]tetrakis[ω-methoxy- (CA INDEX NAME)

RN 362060-09-1 HCAPLUS

CN Cyclotetrasiloxane, octamethyl-, polymer with α, α', α'', α'''-[(2, 4, 6, 8tetramethylcyclotetrasiloxane-2, 4, 6, 8-tetrayl)tetra-3, 1propanediyl]tetrakis[ω-methoxypoly(oxy-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 362060-08-0 CMF (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n C20 H48 O8 Si4 CCI PMS



PAGE 1-B

CM 2

CRN 556-67-2 CMF C8 H24 O4 Si4

RN 362060-10-4 HCAPLUS

CN Cyclotetrasiloxane, octamethyl-, polymer with 2,4,6,8-tetrahexadecyl-2,4,6,8-tetramethylcyclotetrasiloxane and $\alpha,\alpha',\alpha'',\alpha'''-[(2,4,6,8-tetramethylcyclotetrasiloxane-2,4,6,8-tetrayl)tetra-3,1-propanediyl]tetrakis[<math>\alpha$ -methoxypoly(oxy-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 362060-08-0

CMF (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n C20 H48 O8 Si4

CCI PMS

$$\begin{array}{c} \text{PAGE 1-A} \\ \text{MeO} \\ \hline \\ \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{O} \\ \hline \\ \text{In} \\ \end{array} \begin{array}{c} \text{(CH}_2) \\ \text{3} \\ \hline \\ \text{Me} \\ \hline \\ \text{(CH}_2) \\ \text{3} \\ \hline \\ \text{(CH}_2) \\ \text{(CH}_2)$$

PAGE 1-B

CM 2

CRN 362060-06-8 CMF C68 H144 O4 Si4

CM 3

CRN 556-67-2 CMF C8 H24 O4 Si4

IC ICM C08G077-08

INCL 528014000

35-7 (Chemistry of Synthetic High Polymers)

362060-05-7P 362060-07-9P 362060-08-0P

362060-09-1P 362060-10-4P

(polymerization of cyclosiloxanes, (co)polymer cyclosiloxanes, and

(co)polymer cyclosiloxane-oxyalkylenes)

THERE ARE 14 CITED REFERENCES AVAILABLE FOR REFERENCE COUNT: 14 THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L22 ANSWER 17 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2001:237861 HCAPLUS Full-text

DOCUMENT NUMBER: 134:267031

TITLE:

Hydrosilylation curing agents with good

compatibility with unsaturated compounds, their manufacture and uses

INVENTOR(S): Okai, Jiro; Ouchi, Katsuya

PATENT ASSIGNEE(S): Kanegafuchi Chemical Industry Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkvo Koho, 14 pp. CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2001089485	A	20010403	JP 1999-269113	19990922
			<	
PRIORITY APPLN. INFO.:			JP 1999-269113	19990922
			/	

- D Entered STN: 04 Apr 2001
- AB The agents are oligoesters having terminal groups derived from SiH groupcontaining siloxane compds. and are manufactured by hydrosilylating an alkenyl
 group-terminated oligoester precursor with a siloxane compound bearing
 multiple SiH groups. Thus, heating di-Me terephthalate 194 with 2(allyloxy)ethanol 102 and propylene glycol 38 g while removing MeOH gave an
 oligoester diallyloxyethyl ether, which was mixed with KF 9902
 (tetramethylcyclotetrasiloxane) in the presence of Pt complex to give a curing
 agent.
- IT 332016-71-4P, Dimethyl terephthalate-propylene glycol copolymer sru, bis(tetramethylcyclotetrasiloxylpropyl) ester (curing agents; hydrosilylation curing agents with good compatibility with unsatd. compds., manufacture and uses)
- RN 332016-71-4 HCAPLUS
- CN Poly[oxy(methyl-1,2-ethanediyl)oxycarbonyl-1,4-phenylenecarbonyl], $\alpha-[4-[[2-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-yl)propoxy]ethoxy[carbonyl]benzoyl]-<math>\alpha-[2-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-yl)propoxy]ethoxy]- (9CI) (CA INDEX NAME)$

PAGE 1-A

PAGE 1-B

PAGE 1-C

IC ICM C07F007-08

CC 37-2 (Plastics Manufacture and Processing)

T 332015-09-5DP, hydrosilylation products with

tetramethylcyclotetrasiloxane 332016-71-4P, Dimethyl

terephthalate-propylene glycol copolymer sru,

bis(tetramethylcyclotetrasiloxylpropyl) ester

(curing agents; hydrosilylation curing agents with good

compatibility with unsatd. compds., manufacture and uses)

L22 ANSWER 18 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2000:150564 HCAPLUS Full-text

DOCUMENT NUMBER: 132:195541

TITLE: Adhesive polyorganosiloxane compositions rapidly

curable at relatively low temperature

INVENTOR(S): Ono, Kazuhisa

PATENT ASSIGNEE(S): GE Toshiba Silicone Co., Ltd., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 15 pp.

SOURCE: Jpn. Kokai Tokkyo Koho, 15 pp.

DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2000073041	A	20000307	JP 1998-246729	19980901
			<	
PRIORITY APPLN. INFO.:			JP 1998-246729	19980901
			<	

ED Entered STN: 07 Mar 2000

AB The compns., useful for potting, sealing, and junction coating of electronic devices, contain (A) polyorganosiloxanes having ≥2 alkenyl groups bonded to Si atoms, (B) polyorganohydrogensiloxanes having ≥3 Si-H bonds, (C) Pt-based catalysts, and (D) unsatd. group-containing β-diketones RICOCHR2COR4 (R1 = hydrocarbyl; R2 = H, alkyl, AcO; R3, R4 = alkenyl) and/or organosilicon compds. having ≥1 Si-H bond(s) and β-diketone structures RICOCHR2COQ1, RICOCHR2CO2Q2, and/or RSCOCCHR2COQ2Q2 (R1, R2 = same as above; R5 = alkyl; Q1 = C2-6 alkylene; Q2 = C3-6 alkylene). Thus, 100 parts composition containing dimethylvinylsily1-terminated polydimethylsiloxane 100, trimethylsily1-terminated polydimethylsiloxane 100, trimethylsily1-terminated polydimethylsiloxane; isopropanol solution containing 18% chloroplatinic acid 0.05, SiO2 50, and TiO2 5 parts was kneaded with 0.64 part allyl acetoacetate to give a composition showing shear adhesion to Al (120°, 60 min) and poly(butylene terephthalate) (100°, 60 min) of 19 and 11 kq/m2, resp.

IT 259209-18-2DP, trimethylsilyl-terminated (rapidly curable adhesive polyorganosiloxane compns. containing β-diketones for potting, sealing, and junction coating of

electronic devices)

RN 259809-18-2 HCAPLUS

CN Acetic acid, (acetyloxy)-, 3-[2,4,6,8-tetramethyl-4-[2-(tri-2,4cyclopentadien-1-ylsilyl)ethyl]cyclotetrasiloxan-2-yl]propyl ester, polymer with α-(ethenyldimethylsilyl)-ω-[(ethenvldimethvlsilvl)oxv]polv[oxv(dimethvlsilvlene)] and methylsilanediol (9CI) (CA INDEX NAME)

CM 1

CRN 259809-17-1 CMF C28 H44 O8 S15

CM 2

CRN 59942-04-0

CMF (C2 H6 O Si)n C8 H18 O Si2

CCI PMS

$$\begin{array}{c|c} \text{H}_2\text{C} & \text{CH} & \text{Si} & \text{Me} \\ \text{Me} & \text{C} & \text{Me} & \text{Ne} \\ \text{Me} & \text{Me} & \text{Ne} & \text{Ne} \\ \end{array}$$

CM 3

CRN 43641-90-3 CMF C H6 O2 Si

ICM C09J183-05 TC

ICS C09J011-06; C09J183-07

38-3 (Plastics Fabrication and Uses) Section cross-reference(s): 39, 42, 76

- ST polyorganosiloxane adhesive diketone potting sealant coating; electronic packaging silicone rubber adhesive diketone
- IT Adhesives

(cold-curable; rapidly curable adhesive polyorganosiloxane compns. containing β -diketones for potting, sealing, and junction coating of electronic devices)

- IT Coating materials
 - (low-temperature-curable; rapidly curable adhesive polyorganosiloxane compns. containing β -diketones for potting, sealing, and junction coating of electronic devices)
- IT Electronic packaging materials

Potting compositions (rapidly curable adhesive polyorganosiloxane compns. containing β -diketones for potting, sealing, and junction coating of electronic devices)

- IT Silicone rubber, uses
 - (rapidly curable adhesive polyorganosiloxane compns. containing β -diketones for potting, sealing, and junction coating of electronic devices)
- IT Metal alkoxides

(rapidly curable adhesive polyorganosiloxane compns. containing β -diketones for potting, sealing, and junction coating of electronic devices

- IT Silicone rubber, uses
 - (silicate-, rapidly curable adhesive polyorganosiloxane compns. containing B-diketones for potting, sealing, and junction coating of electronic devices)
 - 1118-84-9DP, Allyl acetoacetate, polymers with vinyl-containing siloxanes and trimethylsilyl-terminated Me hydrogen, di-Me siloxane 31900-57-9DP, Dimethylsilanediol homopolymer, dimethylvinylsilylterminated, polymers with vinyl-containing siloxane, trimethylsilylterminated Me hydrogen, di-Me siloxane, and β -diketone-containing alkenes or siloxanes 59942-04-0DP, Dimethylvinylsilyl-terminated polydimethylsiloxane, polymers with vinyl-containing siloxane, trimethylsilyl-terminated Me hydrogen, di-Me siloxane, and β-diketone-containing alkenes or siloxanes 156118-35-3DP. Dimethylsilanediol-methylsilanediol copolymer, trimethylsilylterminated, polymers with vinyl-containing siloxanes and β-diketone-containing alkenes or siloxanes 157550-37-3DP, trimethylsilyl derivs., polymers with vinyl-terminated dimethylsiloxane, trimethylsilyl-terminated Me hydrogen, di-Me siloxane, and β-diketone-containing alkenes or siloxanes 259809-07-9DP, trimethylsilyl-terminated 259809-08-0DP, polymers with vinyl-containing siloxanes and trimethylsilyl-terminated Me hydrogen, 259809-09-1DP, polymers with vinyl-containing siloxanes di-Me siloxane and trimethylsilyl-terminated Me hydrogen, di-Me siloxane 259809-10-4DP, trimethylsilyl-terminated 259809-12-6DP, trimethylsilyl-terminated 259809-14-8DP, trimethylsilyl-terminated 259809-16-0DP, trimethylsilyl-terminated 259809-18-2DP, trimethylsilyl-terminated 259809-20-6DP, trimethylsilyl-terminated (rapidly curable adhesive polyorganosiloxane compns. containing β-diketones for potting, sealing, and junction coating of electronic devices)
- IIT 546-68-9, Titanium tetraisopropoxide 7429-90-5D, Aluminum,
 alkoxides, uses 7440-32-6D, Titanium, alkoxides, uses 7440-67-7D,
 Zirconium, alkoxides, uses

(rapidly curable adhesive polyorganosiloxane compns. containing $\beta\text{--diketones}$ for potting, sealing, and junction coating of

electronic devices)

L22 ANSWER 19 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 1998:352888 HCAPLUS Full-text DOCUMENT NUMBER: 129:58843

TITLE:

Hardenable compositions based on epoxides of cyclic compounds

Weinmann, Wolfgang; Gasser, Oswald; Guggenberger, Rainer; Lechner, Gunther; Soglowek, Wolfgang;

Zech, Joachim

Thera Patent GmbH & Co. KG Gesellschaft fuer PATENT ASSIGNEE(S): Industrielle Schutzrechte, Germany; Weinmann,

Wolfgang; Gasser, Oswald; Guggenberger, Rainer; Lechner, Gunther; Soglowek, Wolfgang; Zech,

Joachim SOURCE:

PCT Int. Appl., 52 pp. CODEN: PIXXD2

Pat.ent.

DOCUMENT TYPE: LANGUAGE: German FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

INVENTOR(S):

	TENT I							API						DATE
										EP65				19971121
			BE,	JP, DE,		ES,	FI,	FR, G	B, GR,	IE,	IT,	LU,	мC	, NL,
DE	1964			A1		1998	0528	DE	1996-		3283			19961121
CA	2246	163		A1		1998	0528	CA	1997-	2246	163			19971121
AU	9854	849		A		1998	0610	AU	1998-	54849	9			19971121
	7193 8792			B2 A1		2000 1998		EP		9512	56			19971121
	8792 R: 1209	AT,			GB,		NL		1997-		9			19971121
CN JP	1106 2001	417 5131	17	B T		2003	0423 0828	JP	1998-		26			19971121
	4001 2155			B2 T		2007 2002			1997-	9512	56			19971121
US	6245	828		В1		2001	0612	US	1998-	10186	57			19980921
US	2002	0022	12	A1		2002	0103	US	2001-		55			20010116
US	2004	0241	13	A1		2004	0205	US	2003-		14			20030318
US PRIORIT	6908 APP			B2		2005	0621				3283	i	A	19961121
								WO	1997-	 EP650	04	1	rī.	19971121
								US			57	i	A 3	19980921

US 2001-759455 B3 20010116 <--

ED Entered STN: 11 Jun 1998

GΙ

CN

AB Hardenable compns., especially useful for dental materials, contain (a) 3-80% ≥1 epoxide having ≥1 epoxycyclohexane or epoxybicycloheptane group, (b) 0-80% ≥1 epoxide different from (a), (c) 3-85% filler, (d) 0.01-25% initiators, inhibitors and/or accelerators, (e) 0-25% auxiliary agents. A typical paste contained diepoxide I 18, 1,1,3,3-tetramethyl-1,3-bis(ethanediyl-3,4epoxycyclohexyl)disiloxane 10, silanized, pigmented quartz 68.9, $(\eta-6$ cumene) (n-5-cvclopentadienvl)iron hexafluorophosphate 1.2, and cumene hydroperoxide 1.8 parts. 208462-90-22 ΙT

(hardenable compns. based on epoxides of cyclic compds. for dental materials)

208462-90-2 HCAPLUS RN

7-Oxabicyclo[4.1.0]heptane, 3.3',3'',3''',3'''',3'''''-[[11-ethyl-11-[2-[3-[3-[2,4,6,8-tetramethyl-4,6,8-tris[2-(7-oxabicyclo[4.1.0]hept-3yl)ethyl]cyclotetrasiloxan-2-yl]propoxy]ethoxy]ethoxy]-4,7,10,12,15,18hexaoxaheneicosane-1,21-diyl]bis[(2,4,6,8tetramethylcyclotetrasiloxane-8, 2, 4, 6-tetrayl)tri-2, 1ethanediyl]]hexakis- (9CI) (CA INDEX NAME)

PAGE 1-A

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PAGE 2-B

PAGE 2-C

PAGE 3-B

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TC
    ICM C08G059-02
    ICS C08G059-22; C08G059-24; C08G059-26; C08G059-32; C08G059-38;
         A61K006-087; C09J163-00
CC
    63-7 (Pharmaceuticals)
    Section cross-reference(s): 37
    208462-64-0P 208462-65-1P 208462-66-2P 208462-67-3P
    208462-70-8P 208462-73-1P 208462-74-2P 208462-75-3P
    208462-76-4P 208462-77-5P 208462-79-7P 208462-82-2P
    208462-83-3P 208462-84-4P 208462-85-5P 208462-86-6P
    208462-87-7P 208462-88-8P 208462-89-9P 208462-90-2P
    208462-91-3P 208462-92-4P 208462-93-5P 208592-39-6P
    208592-41-0P 208592-42-1P 208592-43-2P
                                              208592-44-3P
    208592-45-4P 208592-46-5P
       (hardenable compns. based on epoxides of cyclic compds. for dental
       materials)
REFERENCE COUNT:
                       3
                             THERE ARE 3 CITED REFERENCES AVAILABLE FOR
                             THIS RECORD. ALL CITATIONS AVAILABLE IN THE
                             RE FORMAT
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L22 ANSWER 20 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1995:1005430 HCAPLUS Full-text

DOCUMENT NUMBER: 124:87856
TITLE: Synthesis and Photopolymerization of 1-Propenyl

Ether Functional Siloxanes

AUTHOR(S): Crivello, J. V.; Lohden, G.

CORPORATE SOURCE: Department of Chemistry, Rensselaer Polytechnic

Institute, Troy, NY, 12180, USA

SOURCE: Chemistry of Materials (1996), 8(1),

209-18

CODEN: CMATEX; ISSN: 0897-4756

PUBLISHER: American Chemical Society
DOCUMENT TYPE: Journal

LANGUAGE: English

ED Entered STN: 29 Dec 1995

- AB A variety of mono-, di-, and multifunctional 1-propenyl ether functional siloxanes were readily prepared in high yields by the transition-metal-catalyzed condensation of ac(1-propenyl) o-vinyl ethers with various linear and cyclic H-functional siloxanes. Under these conditions, hydrosilation takes place regioselectively at the vinyl ether site of the ac(1-propenyl) o-vinyl ether. Using onium salt photoinitiators, these new monomers and oligomers undergo rapid polymerization under the influence of UV light. To study these very fast photopolymns, extensive use of Fourier transform realtime IR spectroscopy was made. Employing this technique, the effects of monomer and photoinitiator structure on the rates of polymerization were studied.
- IT 172425-04-6P 172425-19-3P

(monomer; preparation and photopolymn. of propenyl ether functional siloxanes)

- RN 172425-04-6 HCAPLUS
- CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-tetrakis[2-[2-(1-propenyloxy)ethoxy]=thyl]- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

- RN 172425-19-3 HCAPLUS
- CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-tetrakis[2-[2-[2-(1-propenyloxy)ethoxy]ethoxy]ethyl]- (9CI) (CA INDEX NAME)

PAGE 1-B

—
$$CH_2$$
 — CH_2 —

- IT 172425-31-9P 172425-46-6P
- (preparation and photopolymn. of propenyl ether functional siloxanes) RN $\,$ 172425-31-9 $\,$ HCAPLUS

- CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-tetrakis[2-[2-(1-propenyloxy)ethoxy]ethyl]-, homopolymer (9CI) (CA INDEX NAME)
 - CM 1
 - CRN 172425-04-6
 - CMF C32 H64 O12 Si4

PAGE 1-A

PAGE 1-B

- RN 172425-46-6 HCAPLUS
- CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-tetrakis[2-[2-[2-(1-propenyloxy)ethoxy]ethoxy]ethyl]-, homopolymer (9CI) (CA INDEX NAME)
 - CM 1
 - CRN 172425-19-3
 - CMF C40 H80 O16 Si4

PAGE 1-A

PAGE 1-B

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— CH_2— O— CH_2— CH_2— O— CH_2— CH_2— O— CH—— CH—— CH—— CH
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—— CH_2— O— CH_2— CH_2— O— CH_2— CH_2— O— CH—— CH—— CH—— CH
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35-2 (Chemistry of Synthetic High Polymers)
    156118-35-3DP, reaction products with (propenyloxy) (vinyloxy) alkanes
    172425-00-2P 172425-01-3P 172425-02-4P 172425-03-5P
    172425-04-6P 172425-05-7P 172425-06-8P 172425-07-9P
    172425-08-0P 172425-09-1P 172425-10-4P 172425-11-5P
    172425-12-6P 172425-13-7P 172425-14-8P 172425-15-9P
    172425-16-0P 172425-17-1P 172425-18-2P 172425-19-3P
    172425-20-6P 172425-21-7P 172425-22-8P 172425-23-9P
    172425-24-0P 172425-25-1DP, reaction products with di-Me hydrogen Me
    siloxanes 172425-48-8DP, reaction products with di-Me hydrogen Me
    siloxanes 172425-49-9DP, reaction products with di-Me hydrogen Me
    siloxanes 172425-50-2DP, reaction products with di-Me hydrogen Me
    siloxanes 172425-51-3DP, reaction products with di-Me hydrogen Me
    siloxanes
       (monomer; preparation and photopolymn. of propenyl ether functional
       siloxanes)
    172425-26-2P
                  172425-27-3P
                                172425-28-4P
                                               172425-29-5P
    172425-30-8P 172425-31-9P 172425-32-0P 172425-33-1P
    172425-34-2P 172425-35-3P 172425-36-4P 172425-37-5P
    172425-38-6P 172425-39-7P 172425-40-0P 172425-41-1P
    172425-42-2P 172425-43-3P 172425-44-4P 172425-45-5P
    172425-46-6P 172425-47-7P 172585-74-9P 172585-75-0P
    172585-76-1P 172585-77-2P
       (preparation and photopolymn, of propenyl ether functional siloxanes)
L22 ANSWER 21 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN
ACCESSION NUMBER:
                       1994:9628 HCAPLUS Full-text
DOCUMENT NUMBER:
                       120:9628
TITLE:
                       Curing system involving SiH-containing organic
                       oligomers
AUTHOR(S):
                       Iwahara, T.; Kusakabe, M.; Chiba, M.; Yonezawa, K.
CORPORATE SOURCE:
                       Cent. Res. Lab., Kaneka Corp., Kobe, 652, Japan
SOURCE:
                       Journal of Applied Polymer Science (1993)
                       ), 50(5), 825-33
                       CODEN: JAPNAB: ISSN: 0021-8995
DOCUMENT TYPE:
                       Journal
LANGUAGE:
                       English
ED Entered STN: 08 Jan 1994
AB
    The preparation and properties of a novel curing system, which has been
     derived from a SiH-containing organic oligomer (SO) and an allyloxy end-
     blocked organic oligomer (ASO). A mixture of these oligomers could be
     polymerized-crosslinked via a Pt-catalyzed hydrosilylation reaction, i.e., the
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addition reaction of the SiH bonds of \overline{SO} to the allyl functionality of ASO at >100° yielded a rubbery material in a short period of time. The compatibility of SO with ASO was governed by the oligomer main chain employed. SO was completely miscible with ASO, having the same main chain and, thus, the system

system was systematically investigated in detail to manifest its curing behavior and crosslink structure.

151803-09-7P

(crosslinked, preparation and characterization of) 151803-09-7 HCAPLUS

RN

Poly[oxy(methyl-1,2-ethanediyl)], α , α 'methylenebis[ω-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2yl)propoxy]-, polymer with α,α'-methylenebis[ω-(2-

propenyloxy)poly[oxy(methyl-1,2-ethanediyl)]] (9CI) (CA INDEX NAME)

CM 1

CRN 151780-28-8

CMF (C3 H6 O)n (C3 H6 O)n C15 H44 O10 Si8

CCI IDS. PMS

PAGE 1-A

PAGE 1-B

CM 2

CRN 60585-15-1

CMF (C3 H6 O)n (C3 H6 O)n C7 H12 O2

CCI IDS, PMS

$$\text{H2C} \stackrel{\longleftarrow}{\longleftarrow} \text{CH-CH2-O-} \stackrel{\longleftarrow}{\longleftarrow} \text{(C3H6)-O-} \stackrel{\bigcirc}{\longrightarrow} \text{CH2-} \stackrel{\longleftarrow}{\longleftarrow} \text{O-} \text{(C3H6)} \stackrel{\longrightarrow}{\longrightarrow} \text{D} \text{AGE } 1\text{-A}$$

PAGE 1-B

- CH2-CH-CH2

151780-28-8P

(preparation and polymerization of)

RN 151780-28-8 HCAPLUS

CN Poly[oxy(methyl-1,2-ethanediyl)], α,α' methylenebis[ω-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-

vl)propoxy]- (9CI) (CA INDEX NAME)

PAGE 1-A

Me Sin (CH₂) 3-0 (C3H₆) -0
$$\frac{1}{n}$$
 CH₂

PAGE 1-B

CC 37-6 (Plastics Manufacture and Processing)

Section cross-reference(s): 39

151753-06-9P 151803-09-7P

(crosslinked, preparation and characterization of)

124-04-9DP, Hexanedioic acid, reaction products with cyclosiloxanes 126-30-7DP, Neopentyl glycol, reaction products with cyclosiloxanes 502-44-3DP, Caprolactone, reaction products with cyclosiloxanes 9003-31-0DP, functionally terminated derivs. 60585-15-1P 124355-68-6P 150751-13-6P 151679-50-4P 151753-09-2P

151780-28-8P

(preparation and polymerization of)

L22 ANSWER 22 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1993:604050 HCAPLUS Full-text DOCUMENT NUMBER: 119:204050

TITLE: Synthesis of novel organic oligomers containing Si-H bonds

Iwahara, T.; Kusakabe, M.; Chiba, M.; Yonezawa, K. CORPORATE SOURCE: Res. Inst., Kaneka Corp., Kobe, 652, Japan

SOURCE: Journal of Polymer Science, Part A: Polymer

Chemistry (1993), 31(10), 2617-31 CODEN: JPACEC: ISSN: 0887-624X

DOCUMENT TYPE: Journal LANGUAGE: English

LANGUAGE: Engli: ED Entered STN: 13 Nov 1993

AB Several organic oligomers (A) of weight-average mol. weight 103-104 and

terminated with cyclosiloxane rings were successfully synthesized by Ptcatalyzed partial hydrosilylation of an allyloxy (or an allyl carbonate) endblocked linear organic olioner (B) with 2.4.6.8

blocked linear organic oligomer (B) with 2,4,6,8tetramethylcyclotetrasiloxane (I). 13C-MMR anal. revealed that the cyclic structure of I was retained intact in A. The precursors B could be prepared from hydroxyl-terminated oligomers of polytetramethylene glycol, polypropylene glycol, polycaprolactone, or hydrogenated polyisoprene. The storage stability of A was adversely influenced by Pt catalyst residues. The poor stability was improved by decreasing the amount of the Pt catalyst and/or by adding

IT 120268-87-3P

(preparation and storage stability of)

coordinating compds. and mad A excellent stable.

RN 120268-87-3 HCAPLUS

CN Poly[oxy(methyl-1,2-ethanediyl)], α -[3-(2,4,6,8-

tetramethylcyclotetrasiloxan-2-yl)propyl]-0-[3-(2,4,6,8-

tetramethylcyclotetrasiloxan-2-yl)propoxy]- (9CI) (CA INDEX NAME)

PAGE 1-B

CC 35-8 (Chemistry of Synthetic High Polymers) Section cross-reference(s): 37

IT 139268-87-3P 150751-13-6P 150751-14-7P 150751-16-9P (preparation and storage stability of)

L22 ANSWER 23 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1993:219900 HCAPLUS Full-text DOCUMENT NUMBER: 118:219900

TITLE: Dental impression materials

INVENTOR(S): Hattori, Norikazu; Akamatsu, Yasuo PATENT ASSIGNEE(S): Tokuyama Soda Co., Ltd., Japan SOURCE:

Jpn. Kokai Tokkvo Koho, 9 pp. CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 04293955	A	19921019	JP 1991-81132	19910322
			<	
PRIORITY APPLN. INFO.:			JP 1991-81132	19910322
			<	

ED Entered STN: 29 May 1993

AB A hardening material consists of (1) a polyether having alkenyl at the terminals, (2) a polyether having at the terminal polyorganosiloxane with ≥ 1 SiH group and also having ≥ 2 SiH groups in the mol., (3) a polyhydric alc. fatty acid ester, and $(4) \ge 1$ catalyst selected from the group comprising Pt, platinic acid chloride, and Pt complexes. The material is stable and especially suitable for preparing dental impression.

147319-78-6

(dental impression manufacture with fatty acid esters and)

RN 147319-78-6 HCAPLUS

CN Poly[oxy(methyl-1,2-ethanediyl)], α -2-propenyl- ω -(2propenyloxy)-, polymer with α -[3-(2,4,6,8tetramethylcyclotetrasiloxan-2-v1)propv1]-w-[3-(2,4,6,8tetramethylcyclotetrasiloxan-2-yl)propoxy]poly[oxy(methyl-1,2ethanediyl)] (9CI) (CA INDEX NAME)

СМ 1

CRN 120268-87-3

CMF (C3 H6 O)n C14 H42 O9 Si8

CCI IDS, PMS

PAGE 1-A

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PAGE 1-B

CM

CRN 37273-13-5

CMF (C3 H6 O)n C6 H10 O CCI IDS, PMS

$$H_2C \longrightarrow CH \longrightarrow CH_2 \longrightarrow CH$$

TC. ICM C08L071-02

ACCESSION NUMBER:

ICS A61K006-087; A61K006-10; C08G077-46; C08G081-00; C08K005-11; C08L083-12

63-7 (Pharmaceuticals)

37273-13-5 60120-15-2 147319-78-6

(dental impression manufacture with fatty acid esters and)

L22 ANSWER 24 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN 1993:7185 HCAPLUS Full-text

DOCUMENT NUMBER: 118:7185

TITLE: Preparation of organocyclosiloxane-containing

benzenecarboxylate esters

INVENTOR(S): Kasuya, Akira

PATENT ASSIGNEE(S):

Dow Corning Toray Silicone Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 9 pp.

CODEN: JKXXAF Patent

DOCUMENT TYPE: LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 04159287	A	19920602	JP 1990-283831	19901022
JP 2916241	B2	19990705		
PRIORITY APPLN. INFO.:			JP 1990-283831	19901022

- Entered STN: 10 Jan 1993
- For diagram(s), see printed CA Issue.
 - C6H6-m[CO2(R1)nA]m [I; R1 = alkylene, alkyleneoxyalkylene; A = organocyclosiloxane containing ≥2 Si-H bonds; n = 0, 1; m = 1-4] are prepared by addition of C6H6-m(CO2R2)m (R2 = alkenyl, alkenyloxyalkylene; m = 1-4) with

organocyclosiloxane containing 23 Si-H bonds over Pt group catalysts. A mixture of 100 g cyclotetramer II and 10 g bis(2-allyloxyethyl) terephthalate in MePh was refluxed, a mixture of H20, MePh, and excess II was removed, H2PtCl3- tetramethyldivinyldisiloxane complex was added, and the mixture was refluxed to give 22 g title ester III, which was crosslinked to give an elastomer.

IT 144960-77-5P

(preparation of)

RN 144860-77-5 HCAPLUS CN 1.4-Benzenedicarboxvlic acid, bis[2-[3-(2.4.6.8-

tetramethylcyclotetrasiloxan-2-yl)propoxy]ethyl] ester (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

IC ICM C07F007-21

CC 29-6 (Organometallic and Organometalloidal Compounds)

Section cross-reference(s): 39

T 144860-77-5P

(preparation of)

L22 ANSWER 25 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1992:512754 HCAPLUS Full-text

DOCUMENT NUMBER: 117:112754

TITLE: Rapid-curable siloxane compositions with good

mechanical strength

INVENTOR(S): Senba, Makoto; Kusakabe, Masato; Iwahara, Takanao;

Takahara, Tomoko; Yonezawa, Kazuya

PATENT ASSIGNEE(S): Kanegafuchi Kagaku Kogyo K. K., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 19 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1

ITHITHI MCC. MOH. COOMI.

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 04053873	A	19920221	JP 1990-162284	19900620
JP 3308524	B2	20020729	,	
PRIORITY APPLN. INFO.:			JP 1990-162284	19900620
			Z	

- ED Entered STN: 20 Sep 1992
- AB The title compns. comprise (A) organic polymers having mol. weight 500-50,000 and ≥2 hydrosily1 groups, (B) non-polymeric organic compds. having ≥1 alkenyl group, and (C) microcapsulated hydrosily1ation catalysts. Reacting polyoxypropylene glycol and caustic soda with ally1 chloride, treating the alky1-terminated polyoxypropylene with cyclotetra(We hydrogen siloxane) (LS 8600) and H2PtCl6, and mixing 1 g the resulting polymer, 0.03 µL polystyrene-encapsulated reaction product of H2PtCl6 and tetramethyldivinyldisiloxane, and 0.07 g bisphenol A dially1 ether gave a uniform transparent composition, which had snap-up time (100°) in a gelation tester 160 s.
- IT 120268-87-3
- (siloxane compns., rapid-curable, with good mech. strength)
- RN 120268-87-3 HCAPLUS
- CN Poly[oxy(methyl-1,2-ethanediyl)], a-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-yl)propyl]-o-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-yl)propoxy]- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

- IC ICM C08L101-10
 - ICS C08K005-00; C08K009-10; C08L083-05
- CC 37-6 (Plastics Manufacture and Processing)
 II 142-22-3, RAV 7N 1087-21-4, Diallyl isophthalate 1647-16-1,
- 1,9-Decadiene 2370-88-9D, reaction products with polycaprolactonediol ally chloriformate ester 2937-50-0D, ester with

polycaprolactonediol, reaction products with cyclotetra (Me hydrogen siloxane) 3739-67-1, Bisphenol A diallyl ether 9003-31-0D, diallyl ether, reaction products with cyclotetra (Me hydrogen siloxane) 16887-00-6D, Chloride, products with Epol and cyclotetra (Me hydrogen siloxane) 25248-42-dp, Polyloxy(1-oxo-1,6-hexanediyl)), diol derivs., ester with allyl chloroformate, reaction products with cyclotetra (Me hydrogen siloxane) 61488-62-8D, Allyl methacrylatebutyl acrylate copolymer, reaction products with cyclotetra (Me hydrogen siloxane) 120268-87-3 (siloxane commons., rapid-curable, with good mech, strength)

L22 ANSWER 26 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1992:449937 HCAPLUS Full-text

DOCUMENT NUMBER: 117:49937

TITLE: Hydrosilylation-type crosslinkable compositions with good storage stability and thermal curing

rates

INVENTOR(S): Chiba, Makoto; Iwahara, Takanao; Kusakabe, Masato;

Takahara, Tomoko; Yonezawa, Kazuya Kanegafuchi Chemical Industry Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 16 pp.

CODEN: JKXXAF
DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PR

CN

PATENT ASSIGNEE(S):

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 04023867	A	19920128	JP 1990-129475	19900518
			<	
RIORITY APPLN. INFO.:			JP 1990-129475	19900518
			<	

ED Entered STN: 08 Aug 1992

AB The title penetratable, rubbery compns. useful as sealants, adhesives, potting compns., etc., comprise (A) organic polymers having mol. weight 500-50,000 and ≥2 SiH groups, (B) organic polymers having similar mol. weight and ≥1 alkenyl group, and (C) microencapsulated hydrosilylation catalysts. Thus, adding a mixture of allyl-terminated poly(propylene oxide) (I) in PhMe and EPEC16-EtOH solution, and 1,2-dimethoxyethane to LS-8600 (tetramethylcyclotetrasiloxame), and heating gave a vinyl-saturated hydrosilylation product (II). Crosslinkable composition comprising allyl-terminated I 9.0, II 3.2, and mixed poly(vinyl alc.)-polystyrene-encapsulated R2Fc16- tetramethylvinyl siloxane reaction product 0.14 part showed snap-up time 136, 58 and 30 s at 80, 100, and 120°.

(hydrosilylation-crosslinkable rubbery compns. containing, for penetrating sealants with fast curing rates)

RN 142542-12-9 HCAPLUS

Poly[oxy(methyl-1,2-ethanediyl)], α -[3-(2,4,4,6,6,8,8-heptamethylcyclotetrasiloxan-2-yl)propyl]- ω -[3-(2,4,4,6,6,8,8-heptamethylcyclotetrasiloxan-2-yl)propxy]- [9C1] (CA INDEX NAME)

PAGE 1-B

ICM C08L101-02

CC 37-6 (Plastics Manufacture and Processing) Section cross-reference(s): 38, 39

107-05-1D, Allyl chloride, ether with polymer diols, hydrosilylation products with cyclotetrasiloxanes 556-67-2D, hydrosilylation products with vinyl group-terminated polymers 2937-50-0D, Allyl chloroformate, ether with polymer diols, hydrosilylation products with cyclotetrasiloxanes 24980-41-4D, glycols, diether with allyl chloroformate, hydrosilylation products with cyclotetrasiloxanes 25248-42-4D, Poly[oxy(1-oxo-1,6-hexanediyl)], diol derivs., esters with allyl chloroformate 37273-13-5 61488-62-8D, hydrosilylation products with cyclotetrasiloxanes 142542-12-9

(hydrosilylation-crosslinkable rubbery compns. containing, for penetrating sealants with fast curing rates)

L22 ANSWER 27 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1991:410242 HCAPLUS Full-text

DOCUMENT NUMBER: 115:10242

TITLE: Curable hydrophilic compositions containing

polyethers and siloxanes

INVENTOR(S): Hattori, Norikazu; Urabe, Sunao; Kusuki, Koji

PATENT ASSIGNEE(S): Tokuyama Soda Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkvo Koho, 29 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 03047864	A	19910228	JP 1989-149491	19890614
			<	
DE 4019249	A1	19910801	DE 1990-4019249	19900615

PRIORITY APPLN. INFO.:

JP 1989-84363 A1 19890403 <--JP 1989-149491 A 19890614

<--

ED Entered STN: 12 Jul 1991

AB Compns. giving cured hydrophilic products with reproducible dimensions, useful in dentistry, contain alkenyl group-terminated polyethers, siloxane group-terminated polyethers containing 22 SiH groups(alkenyl group-SiH ratio 0.5-10:1), 1-30% silicone oils, and Pt catalysts. A mixture of CH2:CHCH2O(CH2CHM6)0)102CH2CH:CH2 95, pentapropylene glycol bis [3-(3,5,7-trimetylcyclotetrasiloxan-1-yl)propyl] ether 5, quartz powder 100, di-Me silicone (viscosity 10 cSt) 10, and a PtCL4 complex 0.12 part had curing time 5 min, water contact angle 53%, and dimensional change 0.05%.

IT 12(268-87-3 (blends with alkenyl group-terminated polyoxyalkylenes, curable and hydrophilic)

RN 120268-87-3 HCAPLUS

Poly[oxy(methyl-1,2-ethanediyl)], α -[3-(2,4,6,8-

tetramethylcyclotetrasiloxan-2-yl)propyl]-\omega-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-yl)propoxy]- (9CI) (CA INDEX NAME)

PAGE 1-A

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IC ICM C08L071-02

ICS C08L071-02; C08L083-04

ICA C08G065-32

CC 37-6 (Plastics Manufacture and Processing) Section cross-reference(s): 63

IT 120268-87-3

(blends with alkenyl group-terminated polyoxyalkylenes, curable and hydrophilic)

L22 ANSWER 28 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 1991:165693 HCAPLUS Full-text

DOCUMENT NUMBER: 114:165693

TITLE: Curable silicone compositions

INVENTOR(S): Fujiki, Hironao; Shiono, Mikio; Takahashi,
Toshiaki

PATENT ASSIGNEE(S): Shin-Etsu Chemical Co., Ltd., Japan

SOURCE: Eur. Pat. Appl., 14 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 398745	A2	19901122	EP 1990-305414	19900518
EP 398745	A3	19921014		
EP 398745 R: DE, FR, GB	B1	19960904		
JP 02305857	A	19901219	JP 1989-126025	19890519
JP 05036458	В	19930531		
JP 02305858	A	19901219	JP 1989-126026 <	19890519
JP 05036459	В	19930531		
US 5064891	A	19911112	US 1990-525047	19900518
PRIORITY APPLN. INFO.:			JP 1989-126025 A	19890519
			JP 1989-126026 A	19890519

ED Entered STN: 03 May 1991

AB The title compns. giving cured products with good hydrophilic properties and dimensional stability, especially suitable for use as a dental impression material (no data), comprises a curable organopolysiloxane containing ≥2 Sibonded aliphatic unsatd. hydrocarbon radical (Si-auhr), crosslinker, Pt catalyst, and nonionic surfactant of compound having ≥1 Si-H or Si-auhr or a compound having ≥1 hydrophilic polyol and hydrophilic silicone. Thus, a composition containing di-Me vinylsiloxy-endbblocked Me vinyl polysiloxane 100, Me H polysiloxane 2, and polyethylene glycol-modified Me H siloxane surfactant 1 part and 1% iso-PrOH solution of chloroplatinic acid was applied onto a glass plate and cured at room temperature for 10 min to give a product with water contact angle 65°, vs. 93° for a composition not containing the surfactant.

IT 133080-91-8

⁽films, curable, hydrophilic)

RN 133080-91-8 HCAPLUS

CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2-(4,7,10,13-tetraoxatetradec-1-yl)- (CA INDEX NAME)

IC ICM C08L083-07

ICI C08L083-07, C08L083-12

CC 37-6 (Plastics Manufacture and Processing)

IT 112-27-6D, reaction products with siloxanes 4262-92-4 133080-91-8

(films, curable, hydrophilic)

L22 ANSWER 29 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 1991:108985 HCAPLUS Full-text
DOCUMENT NUMBER: 114:108985

TITLE: Polyether siloxanes as dental impression materials

INVENTOR(S): Jochum, Peter; Gasser, Oswald; Zahler, Wolf Dietrich; Lechner, Guenther; Guggenberger, Rainer;

Ellrich, Klaus

PATENT ASSIGNEE(S): ESPE Stiftung und Co. Produktions- und Vertriebs

K.-G., Germany Ger. Offen., 10 pp.

SOURCE: Ger. Offen., 10 :
CODEN: GWXXBX

DOCUMENT TYPE: Patent

DOCUMENT TYPE: Patent
LANGUAGE: German
FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

KIND PATENT NO. DATE APPLICATION NO. DATE DE 3838587 A1 19900517 DE 1988-3838587 19881114 <--CA 2002807 A1 19900514 CA 1989-2002807 19891110 <--CA 2002807 C 19990525 AU 8944602 A 19900517 AU 1989-44602 19891110 <--AU 624192 B2 19920604 JP 02209957 A 19900821 JP 1989-296014 19891113 <--TP 2881586 B2 19990412 EP 1989-121056 EP 369394 A2 19900523 19891114 <--EP 369394 A3 19910403 B1 19930811 R: AT, BE, CH, DE, ES, FR, GB, IT, LI, NL, SE US 5086148 19920204 US 1989-435994 Α 19891114 AT 92746 T 19930815 AT 1989-121056 19891114 <--ES 2058446 T3 19941101 ES 1989-121056 19891114 <---PRIORITY APPLN. INFO.: DE 1988-3838587 A 19881114

<--EP 1989-121056 A 19891114

ED Entered STN: 23 Mar 1991

GI

AB A dental impression material is made of polyether(s), a SiH component, and a Pt catalyst. The polyether has ≥2 vinyl and/or allylene end group, optionally substituted. The SiH component is the reaction product of at least bifunctional allyl or vinyl hydrocarbons comprising ≥1 aromatic, heterocyclic or cycloaliph. ring with the silanes HSIRRI(OSIR2R3)eOSIR4RSR6 or I (R - R6 = H, Me, Et; e, g = 0, 1-8; h = 0, 1-4). Bisphenol-A bisallyloxyethyl ether (7.92 g) was reacted with 9.6 g tetramethylcyclotetrasiloxane, at 55°, in the presence of hexachloroplatinic acid (II). The reaction product (0.27 g) was mixed with 1 g polypropylene glycol diallyl ether and 0.2 g 0.18 II solution

in divinyltetramethyldisiloxane, to give an impression material. IT 132484-63-0P 132484-64-1P 132484-65-2P

132535-53-69

(preparation of, as dental impression material)

RN 132484-63-0 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α,α' -[(1-methylethylidene)di-4,1-phenylene]bis[α -[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-yl)propoxy]-, polymer with 1,1'-(1-methylethylidene)bis[4-[2-[2-(2-propenyloxy)ethoxy]ethoxy]benzene] (9CI) (CA INDEX NAME)

CM

CRN 132484-62-9

CMF (C2 H4 O)n (C2 H4 O)n C29 H56 O10 Si8

PAGE 1-B

CM 2

CRN 132484-61-8

CMF C29 H40 O6

PAGE 1-B

RN 132484-64-1 HCAPLUS

Poly(oxy-1,2-ethanediyl), $\alpha,\alpha'-[(1-methylethylidene)di-4,1-phenylene]bis[<math>\alpha-[3-(2,4,6,8-tetramethyloyclotetrasiloxan-2-yl)propoxy]-$, polymer with 1,1'-(1-methylethylidene)bis[4-(2-propenyloxy)benzene] (9CI) (CA INDEX NAME)

CM

CRN 132484-62-9

CMF (C2 H4 O)n (C2 H4 O)n C29 H56 O10 Si8

PAGE 1-B

CM 2

CRN 3739-67-1

CMF C21 H24 O2

RN 132484-65-2 HCAPLUS

CN Poly(oxy-1,2-ethanediy1), α,α'-[(1-methylethylidene)di-4,1phenylene]bis[ω-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2y1)propoxy]-, polymer with 1,2-bis(2-propenyloxy)benzene (9CI) (CA
INDEX NAME)

CM 1

CRN 132484-62-9

CMF (C2 H4 O)n (C2 H4 O)n C29 H56 O10 Si8

PAGE 1-B

CM 2

CRN 4218-87-5

CMF C12 H14 O2

RN 132535-53-6 HCAPLUS

CN Poly[oxy(methyl-1,2-ethanediyl)], α-2-propenyl-ω-(2propenyloxy)-, polymer with α,α'-[(1-methylethylidene)di-4,1-phenylene]bis[ω-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2yl)propoxy]poly(oxy-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 132484-62-9

CMF (C2 H4 O)n (C2 H4 O)n C29 H56 O10 Si8

PAGE 1-B

CM 2

CRN 37273-13-5

CMF (C3 H6 O)n C6 H10 O

CCI IDS, PMS

IC ICM A61K006-10

ICS C08K005-54; C08K003-08; C08K003-10; C08L071-02; A61C009-00

ICA C07F007-08

ICI C08J003-24, C08L071-02, C08K005-54, C08K003-08, C08K003-10;

CO8K003-10, C08K003-16 CC 63-7 (Pharmaceuticals)

Section cross-reference(s): 38

IT 132484-63-0P 132484-64-1P 132484-65-2P 132535-53-6P

(preparation of, as dental impression material)

L22 ANSWER 30 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 1991:82678 HCAPLUS Full-text

DOCUMENT NUMBER: 114:82678

Anionic ring-opening polymerization of TITLE:

cyclotetrasiloxanes with large substituents AUTHOR(S): Fish, Daryle; Khan, Ishrat M.; Smid, Johannes CORPORATE SOURCE: Fac. Chem., State Univ. New York, Syracuse, NY,

13210, USA

SOURCE: Makromolekulare Chemie, Macromolecular Symposia (

1990), 32(Invited Lect. Int. Symp.

Cationic Polym. Relat. Ionic Processes, 9th,

1989), 241-53 CODEN: MCMSES; ISSN: 0258-0322

DOCUMENT TYPE: Journal LANGUAGE: English

Entered STN: 09 Mar 1991 ED

AB Cyclotetrasiloxanes with bulky substituents were prepared by hydrosilylation of 1,3,5,7-tetramethylcyclotetrasiloxane with 1-hexene, allyl glycidyl ether, allyl ether of triethylene qlycol monomethyl ether, or 4'-allyloxymethylbenzo-18-crown-6. The substituted cyclotetrasiloxanes were homopolymd. or copolymd. with octamethylcyclotetrasiloxane in presence of K silanolate to give siloxanes with side chains containing oxirane rings, crown ethers, or oxyethylene groups. About 50-80% polymer was recovered from the polymerization mixts. The copolymers generally contained less of the substituted monomer than the starting monomer feed, but the bulky substituents did not prevent formation of homopolymers or copolymers.

IT 131718-87-1P

(preparation and characterization of) RN

131718-87-1 HCAPLUS CN

Cyclotetrasiloxane, octamethyl-, polymer with 2,4,6,8-tetramethyl-2,4,6,8-tetrakis(4,7,10,13-tetraoxatetradec-1-v1)cvclotetrasiloxane (9CI) (CA INDEX NAME)

CM

CRN 131718-86-0 CMF C44 H96 O20 S14

PAGE 1-B

- (CH2) 3-0-CH2-CH2-O-CH2-CH2-O-CH2-CH2-OMe

CM 2

CRN 556-67-2

CMF C8 H24 O4 Si4

IT 131718-86-0P

(preparation and polymerization of, with octamethylcyclotetrasiloxane)

RN 131718-86-0 HCAPLUS

CN Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-tetrakis(4,7,10,13-tetraoxatetradec-1-yl)- (9CI) (CA INDEX NAME)

PAGE 1-B

-- (CH2) 3-0-CH2-CH2-O-CH2-CH2-O-CH2-CH2-OMe

- CC 35-7 (Chemistry of Synthetic High Polymers)
- IT 129998-42-1P 131718-87-1P 131718-88-2P 131718-90-6P
- (preparation and characterization of)
- L22 ANSWER 31 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 1989:199251 HCAPLUS Full-text

DOCUMENT NUMBER: 110:199251

TITLE: Preparation of prosthesis materials from polymer mixtures containing siloxane-substituted

polyethers, polyethers, and platinum catalysts INVENTOR(S): Hattori, Norikazu; Urabe, Sunao; Kusumoto, Koshi

PATENT ASSIGNEE(S): Tokuyama Soda Co., Ltd., Japan

SOURCE: Ger. Offen., 42 pp.
CODEN: GWXXBX

DOCUMENT TYPE: Patent LANGUAGE: German

LANGUAGE: Germ
FAMILY ACC. NUM. COUNT: 1

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.		DATE
DE 3741575	A1	19880609	DE 1987-3741575		19871208
DE 3741575	C2	19900613			
JP 01138230	A	19890531	JP 1987-303017 <		19871130
JP 06037558	В	19940518			
US 4877854	A	19891031	US 1987-129173 <		19871207
FR 2607819	A1	19880610	FR 1987-17043		19871208
FR 2607819	B1	19921127			
PRIORITY APPLN. INFO.:			JP 1986-290598 <	A	19861208
			JP 1987-196377 <	A	19870807

ED Entered STN: 26 May 1989

AB A hardenable mixture contains a polyether, a siloxane-substituted polyether, siloxanes, di-Me siloxanes, alkyl Me siloxanes, and ≥1 catalysts selected from Pt. H2PtC16, or Pt complexes. A paste containing 95 parts CH2:CHCH2O(CH2CHMeO)102CH2CH:CH2 and 5 parts J(CH2)30(CHMeCH2O)xCH2)31 (J = L = tetramethylcyclotetrasiloxane residue), and 100 parts quartz was mixed with 0.12 parts H2PtC16-1,3-divinyl-1,1,3,3-tetramethylsiloxane complex to give a caoutchouc-like hardened material with a water contact angle of 53° and a compression deformation of 0.35%. Hardening time was 5 min.

IT 120269-05-8

(dental impressions material containing, caoutchouc-like)

RN 120269-05-8 HCAPLUS

CN Cyclotetrasiloxane, 2,2'-(hexamethyl-4,7,10,13,16,19,22-heptaoxapentacosane-1,25-diyl)bis[2,4,6,8-tetramethyl-, polymer with a-2-propenyl-o-(2-propenyloxy)poly (oxy (methyl-1,2-

ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 120269-04-7

CMF C32 H78 O15 Si8

CCI IDS

PAGE 1-A

6 (D1-Me)

CM 2

CRN 37273-13-5

CMF (C3 H6 O)n C6 H10 O

CCI IDS, PMS

IT 120268-85-1P 120268-88-4P 120268-91-9P 120297-75-8P

(preparation and dental impressions material containing siloxanes and)

RN 120268-85-1 HCAPLUS

CN Poly[oxy(methyl-1,2=ethanediy1]], α -(1,1-dimethyl-2-propenyl)- ω -(1,1-dimethyl-2-propenyl)oxy]-, polymer with α -[1,1-dimethyl-3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-

yl)propyl]-ω-[1,1-dimethyl-3-(2,4,6,8-

tetramethylcyclotetrasiloxan-2-yl)propoxy]poly[oxy(methyl-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 120268-84-0

CMF (C3 H6 O)n C18 H50 O9 Si8

CCI IDS, PMS

PAGE 1-A

$$\begin{array}{c} \text{Me} \\ \text{SIR} \\ \text{III.} \\ \text{Order} \\ \text{Me} \\ \text{Me} \\ \text{Order} \\ \text{Me} \\ \text{Order} \\ \text{Me} \\ \text{Order} \\ \text{Me} \\ \text{Order} \\ \text{O$$

PAGE 1-B

CM 2

CRN 120268-83-9

CMF (C3 H6 O)n C10 H18 O

CCI IDS, PMS

RN 120268-88-4 HCAPLUS

CN Poly(oxy(methyl-1,2=ethanediyl)], α-5-hexenyl-ω-(5-hexenyloxy)-, polymer with α-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-yl)propyl]-ω-[3-(2,4,6,8-tetramethylcyclotetrasiloxan-2-yl)propxylpoly(oxy(methyl-1,2-ethanediyl)) (9C1) (CA INDEX NAME)

CM 1

CRN 120268-87-3

CMF (C3 H6 O)n C14 H42 O9 Si8

CCI IDS, PMS

PAGE 1-A

Me Ne Si H CH2) 3-0 (C3H6) -0
$$\frac{1}{\ln}$$
 (CH2) 3-0

PAGE 1-B

CM 2

CRN 120268-86-2 CMF (C3 H6 O)n C12 H22 O

CCI IDS, PMS

$$\text{H}_2\text{C} \longrightarrow \text{CH} \longrightarrow \text{(CH}_2) \, 4 \longrightarrow \text{O} \longrightarrow \text{(C3H}_6) \longrightarrow \text{O} \longrightarrow \text{n} \quad \text{(CH}_2) \, 4 \longrightarrow \text{CH} \longrightarrow \text{CH}_2$$

RN 120268-91-9 HCAPLUS

CN Poly(oxy-1,4-butanediy1), α -2-buteny1- ω -(2-butenyloxy)-, polymer with α -[2-(2,4,6,8-tetramethylcyclotetrasiloxan-2-

yl)ethyl]- ω -[2-(2,4,6,8-tetramethylcyclotetrasiloxan-2-yl)ethoxy]poly[oxy(methyl-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 120268-90-8

CMF (C3 H6 O)n C12 H38 O9 Si8

CCI IDS, PMS

PAGE 1-A

Me
$$S_1$$
 He CH_2 CH_2 CH_2 CH_2 CH_3 CH_4 CH_5 CH_5 CH_5 CH_6 CH_6 CH_6 CH_7 CH_8 CH

PAGE 1-B

CM 2

CRN 120268-89-5

CMF (C4 H8 O)n C8 H14 O

- RN 120297-75-8 HCAPLUS
- CN Poly(oxy(methyl-1,2=thanediyl)], $\alpha,\alpha',\alpha''-1,2,3$ —
 propanetriyltris[ω -(ethenyloxy)-, polymer with $\alpha=[2-(2,4,6,8-\text{tetramethyloxolotetrasiloxan-2-yl)ethyl]-<math>\omega$ [2-(2,4,6,8-\text{tetramethyloxolotetrasiloxan-2-yl)ethoxy]poly(oxy(methyl-1,2-ethanediyl)] (9CI) (CA INDEX NAME)
 - CM 1
 - CRN 120268-90-8
 - CMF (C3 H6 O)n C12 H38 O9 Si8
 - CCI IDS, PMS

PAGE 1-A

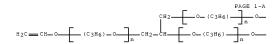
PAGE 1-B

CM 2

CRN 120268-80-6

CMF (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C9 H14 O3

CCI IDS, PMS



PAGE 1-B

____ CH___ CH2

— CH <u>— CH 2</u>

IC ICM C08L071-02

ICS C08L083-10; C08K003-08; C08K003-10; A61K006-10

ICI C08J003-24, C08L071-02, C08L083-10

CC 63-7 (Pharmaceuticals)

IT 120269-05-8

(dental impressions material containing, caoutchouc-like)

120268-97-5P 120268-99-7P

120246-40-4DP, polymers with siloxane-terminated polypropylene glycol

120246-45-9P 120268-85-1P 120268-88-4P 120268-91-9P 120268-94-2P 120268-9

120269-01-4P 120269-03-6P 120297-75-8P 120297-76-9P 120297-78-1P 120297-80-5P 120297-83-8P 120297-85-0P 120297-87-2P 120330-09-8P 120330-11-2P 120376-82-1P 120376-87-6P 120376-89-8P 120376-83-2P 120376-84-3P 120469-12-7P 120516-65-6P (preparation and dental impressions material containing siloxanes and) L22 ANSWER 32 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN 1989:82535 HCAPLUS Full-text ACCESSION NUMBER: DOCUMENT NUMBER: 110:82535 TITLE . Process for manufacture of oxygen-permeable hard contact lens copolymers INVENTOR(S): Mizutani, Yutaka; Harata, Tatsuo; Tanahashi, Naokatsu PATENT ASSIGNEE(S): Nippon Contact Lens, Inc., Japan SOURCE: U.S., 14 pp. CODEN: USXXAM DOCUMENT TYPE: Patent LANGUAGE: English FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE US 4743667 А 19880510 US 1986-888924 19860722 <--JP 1985-193139 A 19850903 PRIORITY APPLN. INFO .: <--Entered STN: 04 Mar 1989 ED GI * STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT * AB O-permeable, solvent-resistant, dimensionally stable, hard contact lens copolymers are prepared by free radical polymerization of monomer mixts. comprising (A) ≥1 multifunctional organosiloxane monomer I and II (A1-A12 = C1-6 alkyl, cycloalkyl, Ph, R2R3R4SiO; R1 = H, Me; R2-R4 = C1-6 alkyl, cycloalkyl, Ph; X1-X7 = C1-6 alkyl, cycloalkyl, Ph, $H2C:CR1CO2CH2CH2CH2CH2CH2CH2; k, k', 1, 1', m, m', n = 0-10), (B) \ge 1$ monofunctional organosiloxane monomer H2C:CR1CO2(CH2)3Si(D)(E)[OSi(A2)(A2')]gA1 and III [R1 = H, Me; A1, A2, A2', A3 = C1-6 alkyl, Ph; D, E = C1-6 alkyl, Ph, R3(SiR2R4O)q; R2-R4 = C1-6 alkyl, Ph;

 ${\bf q}=1-3{\bf l},$ and (C) 21 comonomer selected from C1-10 alkyl, cycloalkyl, or substituted Ph esters of acrylic, methacrylic acids, and itaconate esters and crosslinking agents. Contact lenses manufactured from these copolymers can be easily machined and polished into hard contact lenses. A mixture of 1,3-bis (methacryloxyethoxypropyl-1,1,3,3-pentamethyldisiloxane 20, methacryloxypropyl-1,1,3,3-pentamethyldisiloxane 45, Me methacrylate 22, triethylene glycol dimethacrylate 5, methacrylic acid 8, C.I. Solvent Green 3 0.01 part and 0.01 weight% 2,2'-azobis(2,4- dimethylvaleronitrile) was polymerized at 40° for 20 h to form polymer rods which were molded at 90° for 10 h, and cut into contact lens blanks, and the blanks machined and finished

wetting angle 72.4°, 0-permeability 17.3 + 10-11 (cm2/s) (mL 02/mL-mmHg), no change in color after soaking n-hexane for 30 h, and having 6.1 weight \$ n-hexane content after soaking in n-hexane for 30 h.

IT 118541-96-1P 118541-99-4P

(manufacture of, as oxygen-permeable hard contact lens material)

RN 118541-96-1 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediyl bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate,

(2,4,6,8-tetramethyl-8-propylcyclotetrasiloxane-2,4,6-triyl)tris(3,1-propanediyloxy-2,1-ethanediyl) tris(2-methyl-2-propenoate) and

3-[3,3,3-trimethyl-1,1-bis[(trimethylsilyl)oxy]disiloxanyl]propyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 109479-48-3

CMF C34 H64 O13 S14

PAGE 1-B

--- CH3

CM :

CRN 17096-07-0

CMF C16 H38 O5 Si4

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CM 3
    CRN 97-90-5
    CMF C10 H14 O4
    CM
    CRN 80-62-6
    CMF C5 H8 O2
  H2C
    CM
    CRN 79-41-4
    CMF C4 H6 O2
Me-C-CO2H
    118541-99-4 HCAPLUS
CN
    2-Propenoic acid, 2-methyl-, polymer with cyclohexyl
    2-methyl-2-propenoate, 1,2-ethanediyl bis(2-methyl-2-propenoate),
    methyl 2-methyl-2-propenoate, oxybis(2,1-ethanediyloxy-2,1-ethanediyl)
    bis(2-methyl-2-propenoate), 2-propenoic acid, (1,1,3,3-tetramethyl-1,3-
    disiloxanediyl)bis(3,1-propanediyloxy-2,1-ethanediyl)
    bis(2-methy1-2-propenoate), (2,4,6,8-tetramethy1-8-
    propylcyclotetrasiloxane-2,4,6-triyl)tris(3,1-propanediyloxy-2,1-
    ethanediyl) tris(2-methyl-2-propenoate) and 3-[3,3,3-trimethyl-1,1-
    bis[(trimethylsilyl)oxy]disiloxanyl]propyl 2-methyl-2-propenoate (9CI)
      (CA INDEX NAME)
    CM
        1
    CRN 109479-48-3
    CMF C34 H64 O13 Si4
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PAGE 1-A

PAGE 1-B

CM 4

CRN 109-17-1 CMF C16 H26 O7

PAGE 1-A H2C 0 Me C 0 0 CH2 CH2 0

PAGE 1-B

— Ме

CM 5

CRN 101-43-9 CMF C10 H16 O2

CM 6

CRN 97-90-5

CMF C10 H14 O4

CM 7

CRN 80-62-6 CMF C5 H8 O2



CRN 79-41-4 CMF C4 H6 O2

CM

CRN 79-10-7 CMF C3 H4 O2

IT 109479-48-3P

(preparation and copolymn. of, in oxygen-permeable hard contact lens $\mbox{\tt manufacture})$

RN 109479-48-3 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, (2,4,6,8-tetramethyl-8-propylcyclotetrasiloxane-2,4,6-triyl)tris(3,1-propanediyl)oxy-2,1-ethanediyl) setr (9C1) (CA INDEX NAME)

PAGE 1-B

— CH 3

IC ICM C08F020-22 ICS C08F030-08

INCL 526245000

CC 63-7 (Pharmaceuticals)

Section cross-reference(s): 35 IT 118520-41-5P 118520-42-6P 118520-43-7P 118520-44-8P 118520-45-9P 118520-46-0P 118520-47-1P 118520-48-2P 118520-49-3P 118520-50-6P 118520-51-7P 118520-53-9P 118520-54-0P 118520-55-1P 118520-56-2P 118520-57-3P

118541-96-1P 118541-98-3P 118541-99-4P 118588-77-5P 118611-00-0P

(manufacture of, as oxygen-permeable hard contact lens material) 109456-17-9P 109456-20-4P 109479-46-1P 109479-48-3P ΙT

109479-59-6P

(preparation and copolymn. of, in oxygen-permeable hard contact lens manufacture)

L22 ANSWER 33 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 1989:77390 HCAPLUS Full-text DOCUMENT NUMBER: 110:77390

TITLE:

Silicone-modified polyester resin and silicone-sheathed polyester fibers therefrom

Merrifield, James Hale; Greene, George Homer INVENTOR(S): PATENT ASSIGNEE(S): Union Carbide Corp., USA

SOURCE: Eur. Pat. Appl., 48 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE EP 269023 A2 19880601 EP 1987-117164 19871120 <--EP 269023 A3 19890510 R: AT, BE, DE, FR, GB, IT, NL, SE 19880719 US 1986-933173 19861121 US 4758637 A JP 63165432 A 19880708 JP 1987-292208 19871120 /---C 19910326 CA 1987-552355 CA 1282196 19871120 <--PRIORITY APPLN. INFO.: US 1986-933173 A 19861121 <--US 1985-774962 A2 19850911 <--

ED Entered STN: 04 Mar 1989

AB The title resin comprises a polyester matrix from the reaction of an aromatic dicarboxvlic acid or ester with a diol, a silicone-modified polyester matrix compressing the reaction product of an aromatic dicarboxylic acid or it diester, a diol, and siloxane-block polymer [(I; R = aminoalkyl, aminoaryl, (carboalkoxy)alkyl, (carboaryloxy)alkyl, (carboalkoxy)aryl, (carboarvloxy)arvl, R1 = C1-8 alkyl, arvl, alkenyl, aralkyl; z = alkyl, arvl, aralkyl, alkoxy, polyoxyalkyl, alkenyl, siloxy, and when z = siloxy x = 0, a = 0-10, b = 0-50,000, c = 0-1000, a + b + c is such that polysiloxane block polymer contain \geq Si, x = 0, 1, 2, 3, and optionally a polysiloxane block polymer wherein the silicone-modified polyester matrix is dispersed throughout the polyester matrix or microdomains of 0.05-6 um and contain, when present, encapsulated polysiloxane block polymer. The domains undergo microphase segregation and migration during melt spinning and cold drawing to the surface of the polvester fiber being formed to provide a silicone-sheathed polvester fiber. A drawn polyester fiber prepared from PET precursor and EtO(SiMe2O)200Et containing 5% silicone based on reactants charged had tenacity 2.6 g/denier, 143% elongation, and 23 g/denier modulus, with 86 weight% silicone on the surface.

IT 118859-93-1

(in manufacture of silicone modified polyester fibers)

RN 118859-93-1 HCAPLUS

CN Cyclotetrasiloxane, 2,4,6,8-tetrakis[3-(2-butoxyethoxy)propy1]-2,4,6,8-tetramethyl- (CA INDEX NAME)

PAGE 1-B

___ OBu-n

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TC
    TCM C08G063-68
    ICS D06M015-643; D06M015-507; C08G077-42
    40-2 (Textiles and Fibers)
    556-67-2 1067-99-8
                          7623-01-0 18415-85-5 96322-87-1
ΙT
    97904-71-7 118822-77-8 118859-92-0 118859-93-1
        (in manufacture of silicone modified polyester fibers)
L22 ANSWER 34 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN
                        1988:530143 HCAPLUS Full-text
ACCESSION NUMBER:
DOCUMENT NUMBER:
                        109:130143
TITLE:
                        Polymer electrolyte complexes of lithium
                        perchlorate and comb polymers of siloxane with
                        oligo(oxyethylene) side chains
AUTHOR(S):
                        Fish, Darvle; Khan, Ishrat M.; Wu, E.; Smid,
                        Johannes
CORPORATE SOURCE:
                        Coll. Environ. Sci. For., State Univ. New York,
                        Syracuse, NY, 13210, USA
SOURCE:
                        British Polymer Journal (1988), 20(3),
                        281-8
                        CODEN: BPOJAB; ISSN: 0007-1641
DOCUMENT TYPE:
                        Journal
LANGUAGE:
                        English
ED
    Entered STN: 14 Oct 1988
     Siloxanes with oxyethylene oligomer side chains of the type O(CH2CH2O)7Me and
     (CH2)30(CH2CH2O)nMe (average n ≈7 and 11) were synthesized from di-Me Me H
     siloxane and characterized by 1H and 29Si NMR, IR, and GPC. Cyclic analogs
     were used as model compds. and synthesized from tetramethylcyclotetrasiloxane.
     Polyelectrolyte complexes were made from the comb polymers and LiClO4 by
     solvent-casting from THF, and their ionic conductivities were determined as a
     function of temperature and studied by DSC and correlated with their
     conductivity behavior. Maximum conductivities of .apprx.10-4 S cm-1 were
     achieved at room temperature and ethylene oxide unit-Li+ ratio .apprx.25.
     Crosslinking or blending with high mol. weight poly(oxyethylene) (I) lowered
     the conductance somewhat but vastly improved the mech, properties of the
     complexes, and the blends with I could be best cast into thin, flexible, and
     tough films with good conducting properties.
ΙT
    116614-10-9
        (structure of, as model compds, for comb-branches siloxanes with
       oxyethylene oligomer side chains)
RN
    116614-10-9 HCAPLUS
    Cyclotetrasiloxane, 2,4,6,8-tetramethyl-2,4,6,8-
CN
    tetrakis(4.7.10.13.16.19.22.25-octaoxahexacos-1-v1)- (9CI) (CA INDEX
```

PAGE 1-A

MeO-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-

NAME)

Me O _ CH 2 _ CH 2 _ O _ CH 2

- CC 37-5 (Plastics Manufacture and Processing)
- Section cross-reference(s): 76 IT 556-67-2 2370-88-9 116614-10-9

(structure of, as model compds. for comb-branches siloxanes with oxyethylene oligomer side chains)

L22 ANSWER 35 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1987:583614 HCAPLUS Full-text

DOCUMENT NUMBER: 107:183614

TITLE: Manufacture of polymers for contact lenses

INVENTOR(S): Mizutani, Yutaka; Tanahashi, Naokatsu; Harada,

Tatsuo

PATENT ASSIGNEE(S): Nippon Contact Lens Mfg. Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 9 pp.

DOCUMENT TYPE: CODEN: JKXXAF
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 61285425	A	19861216	JP 1985-126236	19850612
			<	
JP 01060124	В	19891221		
PRIORITY APPLN. INFO.:			JP 1985-126236	19850612
			<	

ED Entered STN: 14 Nov 1987

AB Materials for the manufacture of contact lenses are prepared by copolymg. organosiloxanes, perfluoroalkyl ethers, and vinylcarboxylates (and/or fluoroalkyl alc. vinylcarboxylic acid esters). These materials are wettable, permeable to O, and the lenses prepared from them are worn for an extended period. Thus, a contact lens material was prepared by polymerizing methacryloxyethoxypropylpentamethyldisiloxane 55, F3COCF(CF2)O(CSF60)13CF2CH2OZCCH:CH2 5, Me methacrylate 27, ethylene glycol

F3COCF(CF2)O(C3F6O)13CF2CH2O2CCH:CH2 5, Me methacrylate 27, ethylene glycol dimethacrylate 5, and methacrylic acid 8 parts by weight in the presence of 0.01 part 2,2'-azobis(2,4-dimethylvaleronitrile).

IT 109620-87-3P

(manufacture of, as contact lens materials)

RN 109620-87-3 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, (2,4,6,8-tetramethyl-4,8-dipropylcyclotetrasiloxane-2,6-diyl)bis(3,1-propanediyloxy-2,1-ethanediyl) ester (9C1) (CA INDEX NAME)

PAGE 1-B

- IC ICM G02C007-04
- ICA C08F220-20: C08F220-28: C08F230-08
- CC 63-7 (Pharmaceuticals)
 - Section cross-reference(s): 37

17 79-41-4DP, polymers with acrylates and perfluoropolyoxyalkylenes 80-62-6DP, polymers with acrylates and perfluoropolyoxyalkylenes 88-12-0DP, polymers with acrylates and perfluoropolyoxyalkylenes 97-63-3DP, polymers with acrylates and perfluoropolyoxyalkylenes 97-90-5DP, polymers with acrylates and perfluoropolyoxyalkylenes 101-43-9DP, polymers with acrylates and perfluoropolyoxyalkylenes 109-16-0DP, polymers with acrylates and perfluoropolyoxyalkylenes 109-17-1DP, polymers with acrylates and perfluoropolyoxyalkylenes

352-87-4DP, polymers with acrylates and perfluoropolyoxyalkylenes 617-52-7DP, polymers with acrylates and perfluoropolyoxyalkylenes 868-77-9DP, polymers with acrylates and perfluoropolyoxyalkylenes 923-26-2DP, polymers with acrylates and perfluoropolyoxyalkylenes 1680-21-3DP, polymers with acrylates and perfluoropolyoxyalkylenes 2210-28-8DP, polymers with acrylates and perfluoropolyoxyalkylenes 2998-23-4DP, polymers with acrylates and perfluoropolyoxyalkylenes 3066-71-5DP, polymers with acrylates and perfluoropolyoxyalkylenes 3290-92-4DP, polymers with acrylates and perfluoropolyoxyalkylenes 26248-95-3DP, polymers with acrylates and perfluoropolyoxyalkylenes 84461-14-3DP, polymers with acrylates and perfluoropolyoxyalkylenes 104512-64-3DP, polymers with acrylates and perfluoropolyoxyalkylenes 104534-96-5DP, polymers with acrylates and perfluoropolyoxyalkylenes 109455-83-6DP, polymers with acrylates and perfluoropolyoxyalkylenes 109456-20-4DP, polymers with acrylates and perfluoropolyoxyalkylenes 109620-87-3P 109634-67-5DP, polymers with acrylates and perfluoropolyoxyalkylenes 109635-03-2DP, polymers with acrylates and 109635-05-4P 109635-06-5P perfluoropolyoxyalkylenes 109635-04-3P 109635-07-6P 109635-08-7P 109784-14-7P

(manufacture of, as contact lens materials)

L22 ANSWER 36 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1987:464902 HCAPLUS Full-text

DOCUMENT NUMBER: 107:64902

TITLE: Polymeric materials for contact lens

INVENTOR(S): Mizutani, Yutaka; Harada, Tatsuo; Tanahashi,

PATENT ASSIGNEE(S):

Nippon Contact Lens Mfg. Ltd., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 14 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent. LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 61281216	A	19861211	JP 1985-122539	19850607
			<	
JP 01017128	В	19890329		
PRIORITY APPLN. INFO.:			JP 1985-122539	19850607

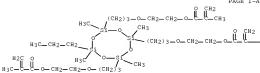
- Entered STN: 21 Aug 1987
- Contact lens are prepared using copolymers of organosiloxanes and acrylic acid AB derivs. The lens are permeable to O and readily wettable. Contact lens prepared from these copolymers can be worn for an extended period. Thus, 1,5bis(methacryloxyethoxypropyl) - 1,1,3,3,5,5-(hexamethyl)trisiloxane 5-80, Me methacrylate 7-82, methacrylic acid 8, triethylene glycol dimethacrylate 5, and 2,2'-azobis(2,4-di-Me valeronitrile) 0.01 part by weight were mixed, placed in a polypropylene cylinder (diameter 16 mm, height 10 mm), and polymerized in N at high temperature to give a transparent rod. Contact lens were prepared from this rod., and the phys. properties were studied.
- 109479-49-4 109479-53-0 109479-55-2 109479-63-2 109479-83-5 109517-09-1 109517-31-9
 - (contact len preparation from)
- 109479-49-4 HCAPLUS
- CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1ethanedivl) bis(2-methvl-2-propenoate), methvl 2-methvl-2-propenoate and (2,4,6,8-tetramethyl-8-propylcyclotetrasiloxane-2,4,6-

triyl)tris(3,1-propanediyloxy-2,1-ethanediyl) tris(2-methyl-2propenoate) (9CI) (CA INDEX NAME)

CM 1

CRN 109479-48-3 CMF C34 H64 O13 Si4

PAGE 1-A



PAGE 1-B

— СН 3

CM 2

CRN 109-16-0 CMF C14 H22 O6

H2C 0 CH2 CH2 - CH

CM 3

CRN 80-62-6

CMF C5 H8 O2

H2C O Me_U_O

CM 4

CRN 79-41-4

CMF C4 H6 O2

Me_U_CO2H

RN 109479-53-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate and (2,4,6,8-tetramethyl-8-propylcyclotetrasiloxane-2,4,6-triylltris(3,1-propanediyloxy-2,1-ethanediyl) tri-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 109479-52-9

CMF C31 H58 O13 Si4

PAGE 1-A

PAGE 1-B

— CH — CH 2

CM 2

CRN 109-16-0

CMF C14 H22 O6

$$\begin{array}{c} ^{\rm H2C} \\ ^{\rm Me} = \begin{array}{c} ^{\rm O} \\ ^{\rm C} = \begin{array}{c} ^{\rm O} \\ ^{\rm C} = \begin{array}{c} ^{\rm O} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ ^{\rm C} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me} = \end{array} \\ \text{Me} = \begin{array}{c} ^{\rm C} \\ \text{Me$$

CM 3

CRN 80-62-6 CMF C5 H8 O2

CM 4

CRN 79-41-4 CMF C4 H6 O2

RN 109479-55-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), (1,1,3,3,5,5-hexamethyl-1,5-trisiloxanediyl) bis(3,1-propanediyloxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate and 2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2-yl)propoxy|ethyl 2-methyl-2-propenoate (SCI) (CA INDEX NAME)

CM 1

CRN 109456-17-9

CMF C24 H48 O8 Si3

$$\begin{array}{c} \text{Me} & \text{Pr-n} \\ \text{S1} & \text{O} & \text{SMe} \\ \text{n-Pr} & \text{Me} \\ \end{array}$$

$$\begin{array}{c} ^{\rm H2C} \\ ^{\rm Me} = \begin{array}{c} ^{\rm O} \\ ^{\rm C} \\ ^{\rm C$$

RN 109479-63-2 HCAPLUS

- CN Butanedioic acid, methylene-, diethyl ester, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate, 2-methyl-2-propenoic acid, (1,1,3,3-tetramethyl-1,3-disiloxanediyl)bis(3,1-propanediyloxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), (2,4,6,8-tetramethyl-8-propylcyclotetrasiloxane-2,4,6-triyl)tris(3,1-propanediyloxy-2,1-ethanediyl) tris(2-methyl-2-propenoate) and 2-13-[3,3,3-trimethyl-1,1-bis[(trimethylsilyl)oxyldisiloxanyl)propoxylethyl 2-methyl-2-propenoate (9C1) (CA INBEX NAME)
 - CM 1
 - CRN 109479-48-3 CMF C34 H64 O13 Si4
 - PAGE 1-A

$$\begin{array}{c} \text{H3C} & \text{CH2} \text{ 3.3.} \\ \text{CH2} \text{ 3.3.} \\ \text{CH2} & \text{CH2} \text{- CH2} \\ \text{CH2} & \text{CH2} \\ \text{CH2} & \text{CH2} \\ \text{CH2} & \text{CH2} \\ \text{CH2} & \text{CH2} \\ \text{CH3} \\ \text{CH3} & \text{CH3} \\ \text{CH3} & \text{CH2} \\ \text{CH2} & \text{CH2} \\ \text{CH2} & \text{CH2} \\ \text{CH2} & \text{CH2} \\ \text{CH2} & \text{CH2} \\ \text{CH3} & \text{CH3} \\ \text{CH3} & \text{CH3$$

PAGE 1-B

- --- CH3
 - CM 2
 - CRN 109456-20-4
 - CMF C22 H42 O7 Si2

$$\begin{array}{c} \text{Mession} \circ \\ \text{Mession} \circ \\ \text{Mession} & \\ \end{array}$$

$$\mathsf{Et} \circ - \overset{\circ}{\mathsf{U}} - \overset{\circ}{\mathsf{U}} - \overset{\circ}{\mathsf{U}} + 2 \\ = \mathsf{CH}_2 - \overset{\circ}{\mathsf{U}} - \circ \mathsf{Et}$$

RN 109479-82-5 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with cyclohexyl 2-methyl-2-propenoate, 1,2-ethanediyl bis(2-methyl-2-propenoate), 1-ethenyl-2-pyrrolidinone, methyl 2-methyl-2-propenoate, oxybis(2,1-ethanediyloxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), [3-[[dimethyl]3-[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxyl propyl]silyl loxy]-1,1,3,5,5-pentamethyl-1,5-trisiloxanediyl]bis(3,1-propanediyloxy-2,1-ethanediyl) bis(2-methyl-2-propenoate) and 2-[3-(2,4,6,8-tripropylcyclotetrasiloxan-2-yl)propoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

1

CRN 109479-46-1 CMF C34 H66 O12 Si4

PAGE 1-B

CM 2

CRN 104858-80-2 CMF C22 H48 O7 Si4

$$\begin{array}{c} \text{Me} & \text{S1} \\ \text{S1} \\ \text{O} \\ \text{S1} \\ \text{O} \\ \text{S2} \\ \text{Ne} \end{array} \right.$$

CM 3

CRN 109-17-1 CMF C16 H26 O7

PAGE 1-B

—_Ме

CM 4

CRN 101-43-9 CMF C10 H16 O2

CM 5

CRN 97-90-5 CMF C10 H14 O4

$$\overset{\text{H2C}}{\text{Me}} \overset{\circ}{\text{U}} \overset{\circ}{\text{U}} \overset{\circ}{\text{U}} \overset{\circ}{\text{CH}_2} \overset{\circ}{\text{CH}_2} \overset{\circ}{\text{CH}_2} \overset{\circ}{\text{U}} \overset{\circ}$$

CN

RN 109517-09-1 HCAPLUS

Butanedioic acid, methylene-, dimethyl ester, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), 1-ethenyl-2-pyrrolidinone, methyl 2-methyl-2-propenoate, oxybis(2,1-ethanediyloxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), (2,4,6,8-tetramethyl-8-propylcyclotetrasiloxane-2,4,6-triyl)tris(3,1-propanediyloxy-2,1-ethanediyl) tris(2-methyl-2-propenoate) and $2-(3-(2,4,6,8-\text{tetramethyl}-4,6,8-\text{tripropylcyclotetrasiloxan-2-yl)propoxylethyl 2-methyl-2-propenoate (9C1) (CA INDEX NAME)$

CM 1

CRN 109479-48-3 CMF C34 H64 O13 Si4

PAGE 1-A

(CH2) 3-O-CH2-CH2-O-CH2-CH3

(CH2) 3-O-CH2-CH2-O-CH2-O-CH2

(CH2) 3-O-CH2-CH2-O-

PAGE 1-B

-- CH3

CRN 104858-80-2 CMF C22 H48 07 Si4

CM 3

CRN 617-52-7

CMF C7 H10 O4

$$\text{MeO-} \overset{\circ}{\text{U}} = \overset{\text{CH}_2}{\text{U}} = \overset{\circ}{\text{CH}_2} = \overset{\circ}{\text{U}} = \text{OMeO}$$

CM 4 CRN 109-17-1 CMF C16 H26 O7 PAGE 1-A H2C 0 Me—U—U—0—CH2—CH2—O—CH2—CH2—O—CH2—CH2—O—CH2—CH2—O PAGE 1-B — Ме CM 5 CRN 109-16-0 CMF C14 H22 O6 CM 6 CRN 88-12-0 CMF C6 H9 N O CM 7 CRN 80-62-6

CMF C5 H8 O2

RN 109517-31-9 HCAPLUS

CN Butanedioic acid, methylene-, diethyl ester, polymer with methyl hydrogen methylenebutanedioate, methyl 2-methyl-2-propenoate, 2-methyl-2-propenoate acid, oxydi-2, l-ethanediyl di-2-propenoate, (2,4,6,8-tetramethyl-8-propylcyclotetrasiloxane-2,4,6-triyl)tris(3,1-propanediyloxy-2,1-ethanediyl) tris(2-methyl-2-propenoate) and 2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2-yl)propxy|ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 109479-48-3

CMF C34 H64 O13 Si4

PAGE 1-A

PAGE 1-B

--- CH3

CM

CRN 104858-80-2

CMF C22 H48 O7 Si4

$$\begin{array}{c} \text{Me} & \text{SI} \longrightarrow 0 \\ \text{SI} \longrightarrow 0 \\ \text{N-Pr} \longrightarrow 0 \\ \text{N-Pr} & \text{Ne} \\ \text{N-Pr} & \text{Me} \\ \end{array}$$

- CM 3
- CRN 4074-88-8 CMF C10 H14 O5
- Hac GH _ CH _ CH 2 _ CH
 - CM 4
 - CRN 2409-52-1
 - CMF C9 H14 O4

- CM 5
- CRN 80-62-6
- CMF C5 H8 O2

- CM 6
- CRN 79-41-4
- CMF C4 H6 O2

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Me_C_CO2H
```

CM 7

CRN 26248-95-3 CMF C6 H8 O4

CCI IDS

CM

CRN 97-65-4 CMF C5 H6 O4

но2С-С-СН2-СО2Н

CM 9

CRN 67-56-1 CMF C H4 O

нзс-он

```
TC
    ICM G02C007-04
    ICS C08F299-08
CC
    63-7 (Pharmaceuticals)
                             109456-22-6 109456-23-7 109456-24-8
IT
    109456-19-1 109456-21-5
    109479-29-0 109479-30-3
                             109479-31-4 109479-32-5 109479-33-6
                             109479-36-9 109479-37-0 109479-38-1
    109479-34-7 109479-35-8
    109479-39-2 109479-41-6
                             109479-43-8 109479-45-0 109479-47-2
    109479-49-4 109479-51-8 109479-53-0 109479-54-1
    109479-55-2 109479-56-3 109479-57-4
                                         109479-58-5
    109479-60-9 109479-61-0
                              109479-62-1 109479-63-2
                             109479-66-5 109479-67-6
    109479-64-3 109479-65-4
                                                      109479-68-7
    109479-69-8 109479-70-1
                             109479-71-2
                                         109479-72-3 109479-73-4
    109479-74-5 109479-75-6 109479-76-7
                                         109479-77-8 109479-78-9
    109479-79-0 109479-80-3 109479-81-4 109479-82-5
    109479-83-6 109479-84-7 109483-75-2 109517-08-0
    109517-09-1 109517-31-9
       (contact len preparation from)
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L22 ANSWER 37 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1987:464901 HCAPLUS Full-text DOCUMENT NUMBER: 107:64901 TITLE:

Preparation of polymeric materials for contact

lens

INVENTOR(S): Mizutani, Yutaka; Harada, Tatsuo; Tanahashi,

Naokats

PATENT ASSIGNEE(S): Nippon Contact Lens Mfg., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 15 pp. CODEN: JKXXAF

DOCUMENT TYPE: Patent
LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 1985-122540 .TP 61281116 19861211 19850607 /--JP 01017129 В 19890329 PRIORITY APPLN. INFO.: TP 1985-122540 19850607 <--

ED Entered STN: 21 Aug 1987

AB Organosiloxanes and acrylic acid derivs. copolymers are prepared for the manufacture of contact lens. The copolymers are permable to 0, having good affinity for H2O. Contact lens prepared from these copolymers are worn for an extended period. Thus, 1-methacryloxyethoxypropyl-3- (acetoxypropyl)-1,1,3,3- (tetramethyl)disiloxane 5-80, Me methacrylate 7-82, methacrylic acid 8, triethylene glycol dimethacrylate 5, 2,2'-azobis(2,4-dimethylvaleronitrile) 0.01 part by weight were mixed, placed in a polypropylene cylinder (diameter 16 mm, height 10 mm), polymerized at high temperature to give a material useful for contact lens preparation.

IT 109455-75-6P 109455-82-5P 109455-95-0P 109478-87-7P 109517-05-7P 109536-12-1P

(preparation of, as contact lens material)

RN 109455-75-6 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), ethyl 2-methyl-2-propenoate and 2-(3-[4,6,8-trisl[2-(acetyloxy)ethyl]-2,4,6,8-tetramethylcyclotetrasiloxan-2-yl]propoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

1

CRN 109455-74-5 CMF C25 H48 O13 Si4

CM 2

CRN 109-16-0 CMF C14 H22 O6

CM 3

CRN 97-63-2

CM ·

CRN 79-41-4 CMF C4 H6 O2

RN 109455-82-5 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 3-(acetyloxy)propyl 2,4,6,8-tetramethyl-6-[3-[2-[(2-methyl-1-oxo-2propenyl)oxy]ethoxy]propyl]-4,8-dipropylcyclotetrasiloxanepropanoate, 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate) and ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 109455-81-4

CMF C27 H54 O11 Si4

PAGE 1-A
$$S_{-}^{Me}$$
 S_{-}^{Fr-n} Me

PAGE 1-B

CM 2

CRN 109-16-0

CMF C14 H22 O6

CM 3

CRN 97-63-2 CMF C6 H10 O2

CM 4

CRN 79-41-4

CMF C4 H6 O2

RN 109455-95-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 2-[3-[3-[3-(acetyloxy)propyl]-1,1,3,3-tertamethyldisiloxanyl]propoxylethyl 2-methyl-2-propenoate, 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), ethyl 2-methyl-2-propenoate and 2-[3-(2,4,6,8-tetramethyl-4,6-8-tripropylcyclotetrasiloxan-2-yl)propxylethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CRN 109455-56-3 CMF C18 H36 O6 Si2

CM 2

CRN 104858-80-2 CMF C22 H48 O7 Si4

$$\begin{array}{c} \text{Pr-n} \\ \text{Si} \longrightarrow \text{O} \\ \text{Ne} \\ \text{O} \longrightarrow \text{Si} \longrightarrow \text{O} \\ \text{O} \longrightarrow \text{CH}_2 \\ \text{O} \longrightarrow \text{Si} \longrightarrow \text{O} \\ \text{O} \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \\ \text{O} \longrightarrow \text{CH}_2 \longrightarrow \text$$

CM 3

CRN 109-16-0

CMF C14 H22 O6

CM 4

CRN 97-63-2 CMF C6 H10 O2

CM 5

CRN 79-41-4 CMF C4 H6 O2

Me_U_CO2H

RN 109478-87-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), ethyl 2-methyl-2-propenoate, 2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2-yl)propoxylethyl 2-methyl-2-propenoate and 2-[3-[4,6,8-tris[2-(acetyloxy)ethyl]-2,4,6,8-tetramethylcyclotetrasiloxan-2-yl)propoxylethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 109455-74-5

CMF C25 H48 O13 Si4

CM 2

CRN 104858-80-2 CMF C22 H48 07 Si4

CMF C22 H48 07 S14

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CM 3
                 CRN 109-16-0
                 CMF C14 H22 O6
   H2C 0 CH2 CH2 CH2 - CH2 
                 CM 4
                 CRN 97-63-2
                 CMF C6 H10 O2
        H2C
                 CM
                 CRN 79-41-4
                 CMF C4 H6 O2
   Me_U_CO2H
RN
                109517-05-7 HCAPLUS
CN
                 2-Propenoic acid, 2-methyl-, polymer with 3-(acetyloxy)propyl
                 2,4,6,8-tetramethv1-6-[3-[2-[(2-methv1-1-oxo-2-
                 propenyl)oxy]ethoxy]propyl]-4,8-dipropylcyclotetrasiloxanepropanoate,
                 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate),
                 methyl 2-methyl-2-propenoate, (2,4,6,8-tetramethyl-8-
                 propylcyclotetrasiloxane-2,4,6-triyl)tris(3,1-propanediyloxy-2,1-
                 ethanediyl) tris(2-methyl-2-propenoate) and 2-[3-(2,4,6,8-tetramethyl-
                 4,6,8-tripropylcyclotetrasiloxan-2-yl)propoxy]ethyl
                 2-methyl-2-propenoate (9CI) (CA INDEX NAME)
                 CM
                              1
                 CRN 109479-48-3
                 CMF C34 H64 O13 Si4
```

PAGE 1-A

PAGE 1-B

PAGE 1-A

PAGE 1-B

CM 3

CRN 104858-80-2

CMF C22 H48 O7 Si4

$$\begin{array}{c} \text{Me} \\ \text{Ne} \\ \text{Ne}$$

CM

CRN 109-16-0 CMF C14 H22 O6

CM 5

CRN 80-62-6

CMF C5 H8 O2

CM 6

CRN 79-41-4

CMF C4 H6 O2

- 109536-12-1 HCAPLUS RN
 - 2-Propenoic acid, 2-methyl-, polymer with 2-[3-[3-[2-(acetyloxy)ethyl]-1,1-bis[[[2-(acetyloxy)ethyl]dimethylsilyl]oxy]-3,3dimethyldisiloxanyl]propoxy]ethyl 2-methyl-2-propenoate, 1,2-ethanedivlbis(oxv-2,1-ethanedivl) bis(2-methyl-2-propenoate),

ethyl 2-methyl-2-propenoate, propyl 2-methyl-2-propenoate and (2,4,6,8-tetramethyl-8-propylcyclotetrasiloxane-2,4,6-triyl)tris(3,1propanediyloxy-2,1-ethanediyl) tris(2-methyl-2-propenoate) (9CI) (CA INDEX NAME)

CM

CRN 109479-48-3

CMF C34 H64 O13 Si4

PAGE 1-B

-- CH3

CM

CRN 109455-59-6 CMF C27 H54 O12 Si4

CM 3

CRN 2210-28-8 CMF C7 H12 O2

H2C o Me_U_U_OPr-n

CM 4

CRN 109-16-0 CMF C14 H22 O6

H2C

CM 5

CRN 97-63-2 CMF C6 H10 O2

H2C O Me_U_U_OEt

CM 6

CRN 79-41-4

CMF C4 H6 O2

Me-C-CO2H

ICM C08F299-08 ICS G02C007-04

CC 63-7 (Pharmaceuticals) IT

109455-57-4P 109455-58-5P 109455-60-9P 109455-61-0P 109455-62-1P 109455-63-2P 109455-64-3P 109455-65-4P 109455-66-5P 109455-67-6P 109455-68-7P 109455-69-8P 109455-71-2P 109455-73-4P 109455-75-6P 109455-76-7P 109455-78-9P 109455-80-3P 109455-82-5P 109455-84-7P 109455-86-9P 109455-88-1P 109455-89-2P 109455-90-5P

109455-91-6P 109455-92-7P 109455-93-8P 109455-94-9P

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10/663.024
    109455-95-0P
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                                  109456-25-9P 109478-82-2P
    109478-83-3P 109478-84-4P
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    109478-88-8P 109478-90-2P
                                 109478-91-3P 109478-92-4P
    109478-93-5P
                  109478-94-6P
                                 109478-95-7P
                                               109478-96-8P
                                               109479-00-7P
    109478-97-9P
                  109478-98-0P
                                 109478-99-1P
    109479-01-8P
                  109479-02-9P
                                 109479-03-0P
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    109479-05-2P
                  109479-06-3P
                                 109479-07-4P
                                               109479-08-5P
    109479-09-6P 109479-10-9P
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    109479-13-2P 109479-14-3P
                                 109479-15-4P 109479-16-5P
    109479-17-6P 109479-18-7P
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    109479-22-3P 109479-23-4P
                                 109479-24-5P 109479-25-6P
    109479-26-7P
                  109479-27-8P
                                 109479-28-9P
                                               109483-76-3P
    109483-77-4P
                   109517-02-4P
                                  109517-03-5P
                                                109517-04-6P
    109517-05-7P
                  109517-06-8P
                                 109517-07-9P
    109536-12-1P
        (preparation of, as contact lens material)
L22 ANSWER 38 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN
ACCESSION NUMBER:
                        1987:107947 HCAPLUS Full-text
DOCUMENT NUMBER:
                        106:107947
ORIGINAL REFERENCE NO.: 106:17595a,17598a
TITLE:
                        Organosiloxane ester copolymers as
                        oxygen-permeable hard contact lens materials
INVENTOR(S):
                        Mizutani, Yutaka; Harata, Tatsuo; Tanahashi,
                        Naokatsu
PATENT ASSIGNEE(S):
                        Nippon Contact Lens Mfg. Ltd., Japan
                        U.S., 9 pp.
                        CODEN: USXXAM
DOCUMENT TYPE:
                        Pat.ent.
```

SOURCE:

LANGUAGE: English FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 4602074	A	19860722	US 1984-682782	19841218
JP 61057612	A	19860324	JP 1984-174177	19840823
JP 05015724	В	19930302		
PRIORITY APPLN. INFO.:			JP 1983-238978 A	19831220
			JP 1984-174177 A	19840823

ED Entered STN: 05 Apr 1987 GT

R4 R2 H2C = CR6CO2CH2CH2O(CH2)3\$i(O\$i)nOmR1 k5 k3 R4 R2 H2C=CR6CO2CH2CH2O(CH2)35i(O5i)nOmk3 TT

A contact lens material is made of organosiloxane ester I and II (R1-R3 = C1-6 alkyl, Ph; R4, R5 = C1-6 alkyl, Ph, R1(SiR2R3O)n; R6 = H, Me; m = 0, 1; n = 1-3) copolymers with an ester of C1-10 monohydroxy or polyhydroxy alc. and an acid selected from methacrylic acid, acrylic acid or itaconic acid. A contact lens was prepared from a copolymer of methacryloxyethoxypropyl tris(trimethylsiloxy)silane 50 and Me methacrylate 37 parts by weight The

hardness of the lens was 78.0 Shore D. the wetting angle 80.6°, and 0 permeability 22.5 + 10-11 mL(STP)cm/cm2.s.mmHq.

104858-81-3P 104858-98-2P 104887-78-7P

ΙT 104887-80-1P 104921-48-4P

(preparation of, as contact lens material)

104858-81-3 HCAPLUS RN

CN 2-Propenoic acid, 2-methyl-, polymer with cyclohexyl 2-methyl-2-propenoate, 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methy1-2-propenoate), methy1 2-methy1-2-propenoate and 2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2vl)propoxylethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 104858-80-2

CMF C22 H48 O7 Si4

CM

CRN 109-16-0

CMF C14 H22 O6

3 CM

CRN 101-43-9

CMF C10 H16 O2

$$\text{CH}_{2}\text{Me}$$

CRN 80-62-6 CMF C5 H8 O2

CM

CRN 79-41-4 CMF C4 H6 O2

RN 104858-98-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with cyclohexyl 2-methyl-2-propenoate, 1, 2-ethanediylbis(oxy-2, 1-ethanediyl) bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate, 2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2yl)propoxy[ethyl 2-methyl-2-propenoate and 2-[3-[3,3,3-trimethyl-1,1bis[(trimethylsilyl)oxy]disiloxanyl]propoxylethyl 2-methyl-2propenoate (9C1) (CA INDEX NAME)

CM 1

CRN 104858-80-2 CMF C22 H48 07 Si4

$$\begin{array}{c} {}^{\rm H2C} \\ {}^{\rm Me} = {}^{\rm U} - {}^{\rm U} - {}^{\rm O} - {}^{\rm CH_2} - {}^{\rm CH_2} - {}^{\rm O} - {}^{\rm CH_2} - {}^{\rm CH_2} - {}^{\rm O} - {}^{\rm CH_2} - {}^{\rm CH_2} - {}^{\rm O} - {}^{\rm CH_2} - {}^{\rm CH_2} - {}^{\rm O} - {}^{\rm CH_2} - {}^{\rm CH_2} - {}^{\rm O} - {}^{\rm CH_2} - {}^{\rm CH_2} - {}^{\rm O} - {}^{\rm CH_2} - {}^{\rm CH_2} - {}^{\rm O} - {}^{\rm CH_2} -$$

$$\underbrace{ \text{OL}_{\text{L}}^{\text{CH}_2} \text{Me} }$$

CRN 79-41-4 CMF C4 H6 O2

RN 104887-78-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate and 2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2-yl)propoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 104858-80-2

CMF C22 H48 O7 Si4

CRN 109-16-0

CMF C14 H22 O6

CM 3

CRN 80-62-6 CMF C5 H8 O2

CPLE CJ no O.



CM 4

CRN 79-41-4 CMF C4 H6 O2

CH2 II Me-II-CO2H

RN 104887-80-1 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), 2-[3-(2,4,4,6,6,8,8-heptamethylcyclotetrasiloxan-2-yl)propoxylethyl 2-methyl-2-propenoate and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 104887-79-8

CMF C16 H36 O7 Si4

CM 2

CRN 109-16-0

CMF C14 H22 O6

CM 3

CRN 80-62-6

CMF C5 H8 O2

CRN 79-41-4 CMF C4 H6 O2

RN 104921-48-4 HCAPLUS

2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), ethyl 2-methyl-2-propenoate, methyl 2-methyl-2-propenoate, 2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2-yl)propoxylethyl 2-methyl-2-propenoate and 2-[3-[3,3,3-trimethyl-1,1-bis[(trimethylsilyl)oxy]disiloxanyl)propoxylethyl 2-methyl-2-propenoate (90)

CM

CRN 104858-80-2 CMF C22 H48 07 Si4

$$\begin{array}{c} \text{Pr-n} \\ \text{Si} \longrightarrow \text{O} \\ \text{Ne} \\ \text{O} \longrightarrow \text{Si} \longrightarrow \text{O} \\ \text{Me} \\ \text{O} \longrightarrow \text{Si} \longrightarrow \text{O} \\ \text{O} \longrightarrow \text{CH}_2 \\ \text{O} \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \\ \text{O} \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \\ \text{O} \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \\ \text{O} \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \longrightarrow \text{CH}_2 \\ \text{O} \longrightarrow \text{CH}_2 \longrightarrow \text$$

CM 2

CRN 104512-64-3 CMF C18 H42 O6 Si4

```
CM 3
                                 CRN 109-16-0
CMF C14 H22 O6
         H2C O CH2 O CH2 O CH2 CH2 O CH
                                 CM 4
                                 CRN 97-63-2
                                   CMF C6 H10 O2
                H2C o
         Me_U_U_OEt
                                 CM 5
                                 CRN 80-62-6
                                   CMF C5 H8 O2
       H2C 0
Me_U_U_OMe
                                 CM 6
                                 CRN 79-41-4
                                   CMF C4 H6 O2
                                 CH2
       Me_U_CO2H
 IC ICM C08F220-26
                                   ICS G02C007-04
 INCL 526245000
CC 63-7 (Pharmaceuticals)
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Section cross-reference(s): 36

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104512-65-4P 104534-97-6P
                                104858-77-7P 104858-78-8P
TT
    104858-81-3P 104858-82-4P
                               104858-87-9P 104858-92-6P
    104858-93-7P 104858-94-8P 104858-95-9P 104858-96-0P
    104858-98-2F 104859-00-9P 104859-02-1P 104887-30-1P
    104887-31-2P 104887-32-3P 104887-34-5P 104887-35-6P
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                                             106855-34-9P
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    106869-65-2P 106869-66-3P 106869-67-4P 106869-70-9P
    106869-71-0P 106869-72-1P 106869-73-2P 106869-74-3P
    106869-75-4P 106869-76-5P 106931-59-3P
```

(preparation of, as contact lens material)

L22 ANSWER 39 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1986:614113 HCAPLUS Full-text

DOCUMENT NUMBER: 105:214113

ORIGINAL REFERENCE NO.: 105:34437a,34440a

TITLE: Contact lens preparation from acrylic polymers

containing siloxanes

INVENTOR(S): Mizutani, Yutaka; Harada, Tatsuo; Tanahashi,

Naokatsu

PATENT ASSIGNEE(S): Nippon Contact Lens Mfg. Ltd., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 11 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 61057612	A	19860324	JP 1984-174177	19840823
JP 05015724	В	19930302	~	
US 4602074	A	19860722	US 1984-682782	19841218
			<	
PRIORITY APPLN. INFO.:			JP 1983-238978 A	19831220
			<	
			JP 1984-174177 A	19840823
			<	

ED Entered STN: 13 Dec 1986

AB Contact lenses are prepared which are hydrophobic and permeable to 0. The lenses are prepared from copolymers consisting of (1) organosiloxanes, (2) alc. esters with methacrylic acid, acrylic acid, itaconic acid, and/or fluoroalkyl alc. and fluoroalkylbenzyl alc. esters with methacrylic, acrylic, or itaconic acids (or just acids themselves). For example, a copolymers was prepared by treating (1) methacryloxyethoxypropyltris(trimethylsiloxy)silane 5-80, (2) di-Me itaconate 7-82, and (3) methacrylic acid 8, and crosslinking agent, triethylene glycol dimethacrylate 5 parts by weight with a polymerization initiator, V-65. The copolymer was molded into a rod and then made into a number of lenses.

IT 104858-81-3 104858-97-1 104858-98-2 104887-78-7 104887-80-1 104921-48-4

(contact lens preparation from)

RN 104858-81-3 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with cyclohexyl 2-methyl-2-propenoate, 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate and 2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2-yl)propxy|ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 104858-80-2

CMF C22 H48 07 Si4

$$\begin{array}{c} \text{Me} & \text{Si} & \text{O} & \text{Me} \\ \text{Si} & \text{O} & \text{Si} & \text{O} & \text{CH}_2 \\ \text{N-Pr} & \text{Si} & \text{O} & \text{SI} & \text{O} & \text{CH}_2 \\ \text{N-Pr} & \text{Me} & \text{O} & \text{CH}_2 & \text{CH}_2 & \text{O} & \text{CH}_2 \\ \text{N-Pr} & \text{Me} & \text{O} & \text{CH}_2 & \text{CH}_2 & \text{O} & \text{CH}_2 \\ \text{N-Pr} & \text{Me} & \text{O} & \text{CH}_2 & \text{CH}_2 & \text{CH}_2 & \text{O} \\ \end{array}$$

CM 2

CRN 109-16-0

CMF C14 H22 O6

CM 3

CRN 101-43-9

CMF C10 H16 O2

CM ·

CRN 80-62-6

CMF C5 H8 O2

CRN 79-41-4 CMF C4 H6 O2

Me_U_C02H

RN 104858-97-1 HCAPLUS

NB Butanedioic acid, methylene-, dimethyl ester, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate, 2-methyl-2-propenoic acid, 2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2yl)propoxylethyl 2-methyl-2-propenoate and 2-[3-[3,3,3-trimethyl-1,1bis[(trimethylsilyl)oxyldisiloxanyl)propoxylethyl 2-methyl-2propenoate (9C1) (CA INDEX NAME)

CM 1

CRN 104858-80-2

CMF C22 H48 O7 S14

CM

CRN 104512-64-3 CMF C18 H42 O6 Si4

CMF C18 H42 O6 Si

$$\begin{array}{c} \text{Me3Si-O} & \text{CH2} \\ \text{Me3Si-O-Si-} & \text{(CH2)} \\ \text{Me3Si-O} & \text{Me3Si-O} \end{array}$$

RN 104858-98-2 HCAPLUS
CN 2-Propenoic acid, 2-methyl-, polymer with cyclohexyl
2-methyl-2-propenoate, 1,2-ethanediylbis(oxy-2,1-ethanediyl)
bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate,

2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2-yl)propoxy]ethyl 2-methyl-2-propenoate and 2-[3-[3,3,3-trimethyl-1,1-bis[(trimethylsilyl)oxy]disiloxanyl]propoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 104858-80-2

CMF C22 H48 O7 Si4

CM 2

CRN 104512-64-3

CMF C18 H42 O6 Si4

CM 3

CRN 109-16-0

CMF C14 H22 O6

CM 4

CRN 101-43-9

CMF C10 H16 O2

CRN 80-62-6 CMF C5 H8 O2

CM

CRN 79-41-4 CMF C4 H6 O2

RN 104887-78-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate and 2-[3-(2,4,6,8-tetramethyl-4,6,8-tripropylcyclotetrasiloxan-2-yl)propoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM :

CRN 104858-80-2

CMF C22 H48 O7 Si4

CRN 109-16-0 CMF C14 H22 O6

H2C O CH2 O

CM 3

CRN 80-62-6 CMF C5 H8 O2

H2C 0

CM 4

CRN 79-41-4 CMF C4 H6 O2

Me_UCH2

RN 104887-80-1 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate), 2-[3-(2,4,4,6,6,8,8-heptamethylcyclotetrasiloxan-2-yl)propoxy]ethyl 2-methyl-2-propenoate and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 104887-79-8

CMF C16 H36 O7 Si4

CM 2 CRN 109-16-0 CMF C14 H22 O6 Me_U_0_CH2_CH2_O_CH2_O_CH2_O_CH2_OH2_OH2_OH2_OH2_OH0 CM 3 CRN 80-62-6 CMF C5 H8 O2 we_U_U_OMe CM CRN 79-41-4 CMF C4 H6 O2 Me_C_CO2H 104921-48-4 HCAPLUS RN CN 2-Propenoic acid, 2-methyl-, polymer with 1,2-ethanediylbis(oxy-2,1ethanediyl) bis(2-methyl-2-propenoate), ethyl 2-methyl-2-propenoate, methyl 2-methyl-2-propenoate, 2-[3-(2,4,6,8-tetramethyl-4,6,8tripropylcyclotetrasiloxan-2-yl)propoxy]ethyl 2-methyl-2-propenoate and 2-[3-[3,3,3-trimethyl-1,1-bis[(trimethylsilyl)oxy]disiloxanyl]prop oxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME) CM 1 CRN 104858-80-2 CMF C22 H48 O7 Si4

$$\begin{array}{c} \text{Me} & \text{Si} \longrightarrow 0 \\ \text{Si} \longrightarrow 0 \\ \text{Si} \longrightarrow 0 \\ \text{Si} \longrightarrow 0 \\ \text{N-Pr} & \text{Si} \longrightarrow 0 \\ \text{N-Pr} & \text{Me} \\ \\ \text{N-Pr} & \text{Me} \\ \end{array}$$

CRN 79-41-4 CMF C4 H6 O2

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ICM C08F230-08
    ICS C08F220-12; C08F220-22; C08F220-40; G02B007-04
CC
    63-7 (Pharmaceuticals)
ΙT
    104512-65-4 104512-66-5
                             104534-97-6 104858-76-6 104858-77-7
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    104858-94-8 104858-95-9
    104858-98-2 104859-00-9
                             104859-02-1 104874-82-0
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    104887-97-0 104921-48-4 104922-92-1 104961-04-8
    105356-19-2
```

(contact lens preparation from)

L22 ANSWER 40 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER. 1981:486337 HCAPLUS Full-text 95:86337

DOCUMENT NUMBER:

ORIGINAL REFERENCE NO.: 95:14497a,14500a

TITLE: Hydrophilic contact lens made from polysiloxanes

containing hydrophilic sidechains

Keogh, Philip L.; Kunzler, Jav F.; Niu, Gregory C.

C. Bausch and Lomb Inc., USA

PATENT ASSIGNEE(S): SOURCE: U.S., 28 pp. CODEN: USXXAM

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

INVENTOR(S):

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 4259467	A	19810331	US 1979-102010	19791210
			<	
CA 1134539	A1	19821026	CA 1980-361551	19801003
			<	

EP	35080	A1	19810909	EP 1980-304295 <		19801128
EP	35080 R: DE, FR, GB,	B1 IT, NI	19840815 SE			
AU	8065231	A	19810618	AU 1980-65231 <		19801210
AU	540143	B2	19841101			
JP	56094324	A	19810730	JP 1980-173364		19801210
				<		
JP	62029777	В	19870629			
PRIORITY	APPLN. INFO.:			US 1979-102010 <	A	19791210

ED Entered STN: 12 May 1984

GT

$$\label{eq:h2C} \begin{array}{c} \text{Me} \text{CMeCO2}\left(\text{CH2}\right) 4 - \begin{array}{c} \text{Me} \\ \text{Sio} \\ \text{Me} \end{array} \begin{array}{c} \text{Me} \\ \text{Sio} \\ \text{Me} \end{array} \begin{array}{c} \text{Me} \\ \text{Si} \\ \text{Me} \end{array} \begin{array}{c} \text{Me} \\ \text{Me} \\ \text{Me} \\ \text{Me} \end{array} \begin{array}{c} \text{Me} \\ \text{Me} \\ \text{Me} \\ \text{Me} \end{array} \begin{array}{c} \text{Me} \\ \text{Me} \\ \text{Me} \\ \text{Me} \\ \text{Me} \end{array} \begin{array}{c} \text{Me} \\ \text{$$

I, R=H

II, R=(CH2)30CH2CH2OCH2CH2OMe

AB Hydrolytically stable, biol. inert, transparent, hydrophilic contact lenses are prepared from a polysiloxane containing hydrophilic sidechains. The structure and preparation of these polysiloxanes are described. Thus, a copolysiloxane (I) was prepared from 1,3-bis(4-methacryloxybutyl)tetramethyldisiloxane [70877-11-1], octamethylcyclotetrasiloxane [556-67-2], and tetramethylcyclotetrasiloxane [2370-88-9]. I was treated with diethylene glycol allyl methyl ether in the presence of H2PtCl6.6H2O in hexane to give II which was a clear colorless fluid, and with 1% diethoxyacetophenone, was molded, the product irradiated with UV to give a cured optically clear hydrophilic contact lens.

T /8142-30-0P

(preparation and reaction of, with methacryloyloxy-terminated siloxanes) 78142-30-0 HCAPLUS

RN 78142-30-0 HCAPLUS CN Cvclotetrasiloxane

Cyclotetrasiloxane, [3-[2-(2-methoxyethoxy)ethoxy]propyl]heptamethyl-(9CI) (CA INDEX NAME)

IC C08F030-08 INCL 526279000

CC 63-7 (Pharmaceuticals)

IT 78142-30-0P

(preparation and reaction of, with methacryloyloxy-terminated siloxanes)

L22 ANSWER 41 OF 41 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1967:412857 HCAPLUS Full-text 67:12857

DOCUMENT NUMBER:

ORIGINAL REFERENCE NO.: 67:2491a

TITLE: Synthesis and properties of siloxane-polyether

copolymer surfactants

AUTHOR(S): Kanner, Bernard; Reid, Wallace George; Petersen,

Ingo H.

CORPORATE SOURCE: Silicones Div., Union Carbide Corp., Tonawanda,

NY, USA

Industrial & Engineering Chemistry Product SOURCE:

Research and Development (1967), 6(2),

CODEN: IEPRA6; ISSN: 0196-4321 DOCUMENT TYPE: Journal

LANGUAGE:

English ED Entered STN: 12 May 1984

Properties of a series of methylsiloxane-oxyalkylene copolymers were compared AB with typical hydrocarbon surface-active agents. The siloxane surfactants were characterized by surface tensions as low as 20 to 21 dynes/cm. in aqueous

solution and relatively small micelles (aggregation nos. of 3.4 and 4.7). As a result of low aqueous surface tensions, certain of these copolymers were excellent wetting agents for low energy hydrophobic surfaces such as polyethylene. Unlike hydrocarbon derivs., methylsiloxane-polyether copolymers

were also surface active in nonaq. polypropylene glycol systems, a contributing factor in the stabilization of polyurethane foam.

17232-95-0

(surface-active)

RN 17232-95-0 HCAPLUS

CN Cyclotetrasiloxane, 2,4,6,8-tetrakis[3-[2-(2-

methoxyethoxy)ethoxy]propy1]-2,4,6,8-tetramethy1- (CA INDEX NAME)

PAGE 1-B

--- CH2--- CH2--- CH2--- CH2--- OMe

CC 46 (Surface Active Agents and Detergents) IT 17065-96-2 17065-97-3 17065-98-4 17065-99-5 17066-00-1 17066-01-2 17170-72-8 17232-95-0 17423-00-6 (surface-active)

=> d his nofile (FILE 'HOME' ENTERED AT 07:21:37 ON 19 MAR 2008) FILE 'REGISTRY' ENTERED AT 07:21:56 ON 19 MAR 2008 ACT ECH024/A L1 STR L2 50 SEA SSS SAM L1 -----L3 STR L1 50 SEA SSS SAM L3 T. 4 33032 SEA SSS FUL L3 L5 SAV L5 ECH024A/A 1.6 STR L1 L7 46 SEA SUB=L5 SSS SAM L6 763 SEA SUB=L5 SSS FUL L6 L8 SAV L8 ECH024B/A FILE 'HCAPLUS' ENTERED AT 07:44:07 ON 19 MAR 2008 L9 402 SEA ABB=ON PLU=ON L8 FILE 'REGISTRY' ENTERED AT 07:44:16 ON 19 MAR 2008 L10 STR L1 L11 10 SEA SUB=L5 SSS SAM L10 L12 167 SEA SUB=L5 SSS FUL L10 SAV L12 ECH024C/A 1.13 96 SEA ABB=ON PLU=ON L12 NOT 1-100/X FILE 'HCAPLUS' ENTERED AT 07:47:00 ON 19 MAR 2008 L14 51 SEA ABB=ON PLU=ON L13 41 SEA ABB=ON PLU=ON L14 AND (1840-2004)/PRY, AY, PY L15 L16 1 SEA ABB=ON PLU=ON L15 AND ELECTROCHEM? 7 SEA ABB=ON PLU=ON L15 AND ELECTRO? L17

41 SEA ABB=ON PLU=ON (L15 OR L16 OR L17) 46 SEA ABB=ON PLU=ON L9 AND ELECTRO?

21 SEA ABB=ON PLU=ON L9 AND ELECTRO?/SC.SX

3 SEA ABB=ON PLU=ON L15 AND ELECTRO?/SC,SX 41 SEA ABB=ON PLU=ON L18 OR

L18

L19

L20

L21 L22